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WITNESS my hand this  
Seventh day of December 2004

A handwritten signature in black ink, appearing to be "L. Mynott".

LEANNE MYNOTT  
MANAGER EXAMINATION SUPPORT  
AND SALES

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**AUSTRALIA**  
**Patents Act 1990**

**The University of Newcastle Research Associates Limited**  
**and**  
**Children's Medical Research Institute**

**PROVISIONAL SPECIFICATION**

*Invention Title:*

*Methods and agents for inhibiting dynamin-mediated  
endocytosis II*

The invention is described in the following statement:

## **Methods and agents for inhibiting dynamin-mediated endocytosis**

### **Field of the Invention**

The present invention relates to the provision of agents for inhibiting dynamin mediated endocytosis and methods for the prophylaxis or therapeutic treatment of diseases or conditions mediated by dynamin-dependent endocytosis.

### **Background of the Invention**

Mammalian cells take up extracellular material and recycle their membranes by endocytosis which involves the formation of numerous membrane vesicles at the plasma membrane. The vesicles occur in different sizes, ranging from large phagosomes, smaller clathrin-coated vesicles to tiny synaptic vesicles (SV). Endocytic mechanisms subserve many cellular functions including the uptake of extracellular nutrients, regulation of cell-surface receptor expression and signalling, antigen presentation and maintenance of synaptic transmission.

Among the various endocytic pathways are two that are biochemically well-characterized. The first is rapid synaptic vesicle endocytosis (SVE) that follows vesicle exocytosis in nerve terminals. SVE is not specifically linked to receptor activation but serves to retrieve empty SVs for later refilling, and requires the activity of the enzyme dynamin I. The second is receptor-mediated endocytosis (RME) which is initiated upon ligand binding to cell surface receptors and occurs via clathrin-coated pits in all cells, including nerve terminals. RME provides the main entry point into cells for plasma membrane components (such as the receptor-ligand complexes and membrane lipids) or for extracellular fluid and involves the action of dynamin II. Both RME and SVE operate together within the same neuron but perform distinct functional roles.

Although they share similar underlying protein machinery, RME and SVE utilise distinct isoforms of the same proteins. Multiple subforms of both RME and SVE exist. For example, internalisation of the epidermal growth factor receptor (EGFR) and transferrin receptors are mediated by RME and are dependent on the activity of dynamin, but only the former is sensitive to tyrosine kinase (TK) inhibitors suggesting distinct biochemical requirements for RME of these two activated receptors. Endocytosis plays multiple roles in

human pathological conditions including neuronal disorders and a better understanding of how to control endocytosis is clinically important.

Dynamin is the key enzyme which mediates the final stage of endocytosis (Brodin et al., 2000). As well as dynamin, the molecular mechanisms of endocytosis involve many proteins and lipid cofactors that result in dynamin recruitment and its activation (Cousin and Robinson., 2001). The endocytic proteins act sequentially in endocytosis at stages which fall into at least 4 morphological and biochemical categories although some proteins are involved at more than a single stage in the pathway. These morphological and biological categories are:

*Nucleation:* The presynaptic terminal synaptotagmin on the SV functions as the link between exocytosis and endocytosis by recruiting the AP-2 adaptor protein complex to nucleation points at sites of exocytosis. AP-2 recruits clathrin to form a vesicle coat and then amphiphysin.

*Invagination:* Amphiphysin is a docking molecule that recruits most of the remaining endocytic proteins (dynamin, endophilin and synaptojanin) required for the vesicle to invaginate.

*Fission:* Rings of assembled dynamin, amphiphysin and/or endophilin form as a helical collar around the neck of invaginated vesicles. All three of these proteins are able to self-assemble into rings *in vitro*. Fission of the vesicle neck leading to release of the vesicle requires the GTPase activity of dynamin. GTP hydrolysis produces sudden expansion of the helix pushing the vesicle from the plasma membrane or alternatively, causes ring constriction. In the *Drosophila* strain *shibire*, mutations in dynamin's GTPase domain (that do not block GTP binding but block GTP hydrolysis) allow assembly of dynamin helices yet block SV fission after they form (Koenig and Ikeda., 1989). This discriminates between the GTP binding and GTP hydrolysis steps of dynamin's reaction cycle and indicates that GTP hydrolysis that is, GTPase activity, is the last step prior to vesicle fission. Overexpression of GTPase-defective dynamin mutants inhibits both RME and SVE (Brodin et al., 2000).

*Uncoating:* The SV is uncoated and filled with neurotransmitter before being available for exocytosis.

Accordingly, dynamin is a GTPase enzyme required for the retrieval of synaptic vesicles after exocytosis and functions in endocytosis by stimulated assembly as a helix

around the neck of invaginating synaptic vesicles (Brodin et al., 2000; Cousin and Robinson., 2001).

Dynamin is also a phosphoprotein and is phosphorylated by protein kinase C (PKC) *in vitro* and by cyclin-dependent protein kinase (Cdk5) *in vivo*. It is rapidly dephosphorylated by calcineurin on stimulation of endocytosis by depolarisation and calcium influx, and blocking dephosphorylation prevents endocytosis in nerve terminals. It remains dephosphorylated during endocytosis of most vesicles and is rephosphorylated while endocytosis is completing. Hence, the dephosphorylation of dynamin is unlikely to play a role during endocytosis but is probably a priming step prior to endocytosis.

There are three dynamin genes with dynamin I being expressed in neurons while dynamin II is ubiquitously expressed. Dynamin III is expressed in neurons and is highly abundant in testes. All dynamins have four main domains namely, the GTPase domain, the pleckstrin homology (PH) domain, the GTPase effector domain (GED), and a proline rich domain (PRD).

The GTPase domain has an unusually low affinity for GTP (10–25  $\mu$ M) and extremely high turnover rates compared with other GTPases. It is required for vesicle fission. The crystal structure of this domain of dynamin from *Dictyostelium* was recently solved (Niemann et al., 2001). The globular structure contains the G-protein core fold, but the normal six-stranded  $\beta$ -sheet is extended to an eight-stranded one by a unique 55 amino acid insertion.

The pleckstrin homology (PH) domain is both a targeting domain and potentially a GTPase inhibitory module and is essential for endocytosis. Dynamin interacts with lipids via this domain, and dynamin binding to nanotubules containing phosphatidylinositol bisphosphate (PtdIns(4,5)P<sub>2</sub>) greatly stimulates GTPase activity (Stowell et al., 1999). The PH domain is not needed for self-assembly or GTPase activity and deleting it (delta-PH dynamin) maximally increases intrinsic GTPase activity.

The GTPase effector domain (GED) controls dynamin-dynamin interactions and dynamin assembly into a tetrameric configuration. About 28–32 tetramers cooperatively self-assemble as a single ring or as a helix around PtdIns(4,5)P<sub>2</sub>-containing lipid mixtures. GED accounts for tetramer self-association by binding to the GTPase domain. Mutations in

GED affect endocytosis in cells, some decreasing and some (surprisingly) increasing endocytosis. GED acts like a GTPase activator protein to stimulate GTPase activity.

The proline-rich domain (PRD) at dynamin's C-terminus interacts with many SH3 domain-containing proteins and calcineurin, and is the site for *in vivo* dynamin phosphorylation.

Multiple endocytosis inhibitors and methods for inhibiting endocytosis exist such as cationic amphiphilic drugs (eg., chlorpromazine), concanavalin A, phenylarsine oxide, dansylcadaverine, intracellular potassium depletion, intracellular acidification and decreasing medium temperature to 4°C. Each has poor specificity and limited utility.

Nonetheless, their use has contributed to a better understanding of endocytosis. Some have been used to demonstrate that blocking endocytosis has clinical implications for humans (Atwood., 2001).

### Summary of the Invention

The present invention in one or more embodiments relates to compounds capable of binding to dynamin and thereby inhibiting the GTPase activity of the protein, as well as the use of such compounds to inhibit dynamin-mediated endocytosis. In particular, at least some dimeric tyrphostins have been found to be capable of inhibiting the GTPase activity of dynamin and thereby, to be useful for inhibiting endocytosis mediated by dynamin.

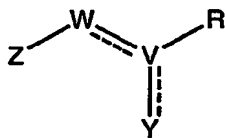
Accordingly, in an aspect of the present invention there is provided a method for inhibiting dynamin-dependent endocytosis in cells, the method comprising administering to the cells an effective amount of a dimeric tyrphostin, or an analogue thereof, which binds to dynamin and thereby inhibits GTPase activity of dynamin, or a physiologically acceptable salt of the dimeric tyrphostin or analogue thereof.

Typically, a compound for inhibiting dynamin-dependent endocytosis in accordance with a preferred embodiment of the invention will be a compound of formula I or a physiologically acceptable salt thereof, wherein:

M-Sp-M'

Formula I

M and M' are each independently a moiety of formula II and are the same or different, and Sp is a spacer;



Formula II

V is C or CH;

5 W is CH or a linker group; and

Y is hydrogen, cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, sulfur, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; or

10 W, V and Y form a 5 or 6 membered substituted or unsubstituted heterocyclic or carbocyclic ring fused with Z, wherein the heterocyclic ring includes from 1 to 3 heteroatoms selected from O, N and S, and the carbocyclic or heterocyclic ring, when substituted, has at least one substituent selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, sulfur, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; and

R is CH<sub>2</sub>R', CXR' or CHX'R';

X is O or S;

20 X' is cyano, nitro, amino, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur;

R' is NH, O or S bonded to the spacer; and

Z is a group selected from:

- 25 (a) an unsubstituted heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S;
- (b) an unsubstituted carbocyclic group consisting of one or two rings independently having 5 or 6 ring members;

(c) a heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S wherein the heterocyclic group has one or more substituents independently selected from:

- (i) nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and
- (ii) a C<sub>1</sub>-C<sub>2</sub> alkyl or C<sub>1</sub>-C<sub>2</sub> alkenyl group with at least one substituent selected from nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and

(d) a carbocyclic group consisting of one or two rings independently having 5 or 6 ring members, and at least two substituents when W is CH or a linker group or W, V and Y form an unsubstituted carbocyclic group, or at least one substituent when W, V and Y form a heterocyclic group, independently selected from:

- (i) nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and
- (ii) a C<sub>1</sub>-C<sub>2</sub> alkyl or C<sub>1</sub>-C<sub>2</sub> alkenyl group with at least one substituent selected from nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl;

wherein when the Z group of one of M or M' is selected from (b), the Z group of the other of M or M' is selected from (a), (c) or (d).

The invention also relates to the prophylaxis or therapeutic treatment of a disease or condition responsive to inhibition of dynamin-dependent endocytosis.

Accordingly, in another aspect of the present invention there is provided a method for prophylaxis or therapeutic treatment of a disease or condition in a mammal mediated by dynamin-dependent endocytosis, the method comprising treating the mammal with an effective amount of a dimeric tyrphostin, or an analogue thereof, which binds to dynamin and thereby inhibits GTPase activity of dynamin, or a physiologically acceptable salt of the dimeric tyrphostin or analogue thereof.

In another aspect of the present invention there is provided a method for prophylaxis or therapeutic treatment of a disease or condition in a mammal mediated by



dynamin-dependent endocytosis, the method comprising treating the mammal with an effective amount of a compound of formula I or a physiologically acceptable salt thereof.

It is not necessary that a compound administered to the mammal in accordance with the invention be in a dimeric form. Hence, "treating" the mammal with a dimeric tyrphostin or compound of formula I is to be taken for instance to encompass administration of compounds that dimerise *in vivo* to produce a dimeric tyrphostin or compound of formula I, and prodrugs which are processed *in vivo* to yield or produce a dimeric tyrphostin or compound of formula I.

In a further aspect of the present invention there is provided the use of a compound of formula I or a physiologically acceptable salt thereof in the manufacture of a medicament for prophylaxis or therapeutic treatment of a disease or condition in a mammal mediated by dynamin-dependent endocytosis.

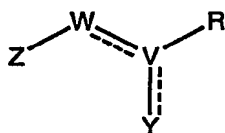
In yet another aspect of the present invention there is provided the use of a dimeric tyrphostin or an analogue thereof, which binds to dynamin and thereby inhibits GTPase activity of dynamin, in the manufacture of a medicament for prophylaxis or therapeutic treatment of a disease or condition in a mammal mediated by dynamin-dependent endocytosis.

In still another aspect of the present invention there is provided a compound of formula III or a physiologically acceptable salt thereof, wherein:



Formula III

M and M' are each independently a moiety of formula IV and are the same or different, and Sp is a spacer.



Formula IV

V is C or CH;

W is CH or a linker group; and

Y is hydrogen, cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, sulfur, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least

one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; or

W, V and Y form a 5 or 6 membered substituted or unsubstituted heterocyclic or carbocyclic ring fused with Z, wherein the heterocyclic ring includes from 1 to 3

- 5 heteroatoms selected from O, N and S, and the carbocyclic or the heterocyclic ring, when substituted, has at least one substituent selected from cyano, NH, nitro, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, sulfur, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; and

10 R is CH<sub>2</sub>R', CXR' or CHX'R';

X is O or S;

X' is cyano, nitro, amino, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, or an unsubstituted C<sub>1</sub>-C<sub>3</sub> group or C<sub>1</sub>-C<sub>3</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl,

15 carboxy, thiocarboxy and sulfur;

R' is NH, O or S bonded to the spacer; and

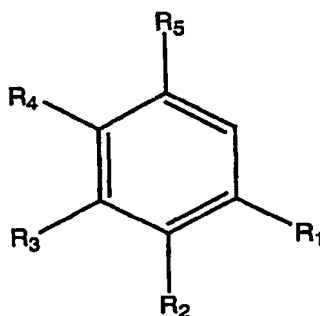
Z is a group selected from:

- (a) an unsubstituted heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S;
- 20 (b) an unsubstituted carbocyclic group consisting of one or two rings independently having 5 or 6 ring members;
- (c) a heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S, wherein the heterocyclic group has one or more substituents independently selected from:
- 25 (i) nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and
- (ii) a C<sub>1</sub>-C<sub>2</sub> alkyl or C<sub>1</sub>-C<sub>2</sub> alkenyl group with at least one substituent selected from nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and
- 30 (d) a carbocyclic group consisting of one or two rings independently having 5 or 6 ring members, and at least two substituents when W is CH or a linker group or W, V and Y form an unsubstituted carbocyclic group, or at least

one substituent when W, V and Y form a heterocyclic group, independently selected from:

- (i) nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; and
- (ii) a C<sub>1</sub>-C<sub>2</sub> alkyl or C<sub>1</sub>-C<sub>2</sub> alkenyl group with at least one substituent selected from nitro, NH, amino, cyano, halo, hydroxy, carboxy, oxo, sulfur, sulfhydryl, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl;

wherein when the Z group of one of M or M' is selected from (b), the Z group of the other of M or M' is selected from (a), (c) or (d), and with the proviso that the Z group of at least one of M and M' is other than a benzene group of formula IVa when R is CXR', X is O, R' is NH bonded to the spacer, V is C, W is CH, Y is cyano, and



Formula IVa

R<sub>1</sub>, R<sub>2</sub>, and R<sub>5</sub> are H, and R<sub>3</sub> and R<sub>4</sub> are hydroxy; or  
 R<sub>1</sub> and R<sub>5</sub> are H, and R<sub>2</sub> to R<sub>4</sub> are hydroxy when Sp is a C<sub>2</sub>-C<sub>4</sub> alkyl spacer.

Preferably, when the Y substituent of one of M or M' of a compound of the invention or administered in accordance with the invention is hydrogen, the Y substituent of the other of M or M' will be other than hydrogen. Typically, the Z group of at least one of M and M' of a compound of the invention or administered in accordance with a method of the invention will be other than a 2,3-disubstituted carbocyclic group. Preferably, the Z group of at least one of M and M' will have:

at least two substituents ortho relative to one another or in adjacent substitution positions when the Z group is selected from (d) and W is CH or a C<sub>1</sub>-C<sub>3</sub> linker group; or

the, or one of, the substituents on a carbon atom adjacent to the, or one of the, heteroatom(s) when the Z group is a heterocyclic group selected from (c); or

5 the, or one of, the substituents on a carbon atom spaced at least one bond length from the heterocyclic ring formed when W, V and Y are cyclised and form this type of group.

Preferably, the linker group of a moiety of formula II or IV will comprise a single atom or a chain of up to three atoms in length wherein the, or one or more of the atoms, may be an atom other than carbon such as N, O or S. Preferably, the linker group will be a C<sub>1</sub>-C<sub>3</sub> linker group. The linker group may be substituted or unsubstituted, and may  
10 include one or more double bonds. Substituents may for instance be selected from hydroxy, amino, halo, nitro or groups which do not substantially adversely impact on the activity of the compound. Most preferably, the linker group will be unsubstituted.

In a further aspect of the present invention there is provided a pharmaceutical composition comprising a compound of formula III, or a physiologically acceptable salt  
15 thereof, together with a physiologically acceptable carrier.

In another aspect of the present invention there is provided a method for screening a dimeric tyrphostin or an analogue thereof for ability to bind to dynamin and inhibit GTPase activity of dynamin, wherein the method comprises:

20 incubating the dimeric tyrphostin or analogue thereof with dynamin or a molecule having dynamin GTPase activity; and

determining whether the dimeric tyrphostin or analogue thereof inhibits the GTPase activity of dynamin.

The molecule having dynamin GTPase activity may be a fragment of dynamin that retains GTPase activity or for instance, a homologue, derivative or analogue of dynamin  
25 that acts as a substitute for dynamin in the assay.

By the term "dimeric tyrphostin" is meant a compound comprising two tyrphostin moieties linked together by a spacer moiety wherein the tyrphostin moieties are the same or different. Typically, each tyrphostin moiety will be the same. Most preferably, each tyrphostin moiety will be a benzyldenemalonitrile moiety. Bis-tyrphostin is one such

dimeric tyrphostin which has now surprisingly been found to be capable of binding to dynamin and inhibiting the GTPase activity of the protein.

All publications mentioned in this specification are herein incorporated by reference. Any discussion of documents, acts materials, devices, articles or the like which has been included in the present specification is solely for the purpose of providing a context for the present invention. It is not to be taken as an admission that any or all of these matters form part of the prior art base or were common general knowledge in the field relevant to the present invention as it existed in Australia or elsewhere before the priority date of this application.

Throughout this specification the word "comprise", or variations such as "comprises" or "comprising" will be understood to imply the inclusion of a stated element, integer or step, or group of elements, integers or steps, but not the exclusion of any other element, integer or step, or group of elements, integers or steps.

The features and advantages of the present invention will become further apparent from the following description of preferred embodiments of the invention together with the accompanying drawings.

### **Brief Description of the Drawings**

Figure 1: Graphs indicating bis-tyrphostin and tyrphostin A47 (a, b) inhibit the GTPase activity of both dynamin I and dynamin II. The GTPase activity of 0.2 g dynamin I (c,d) and dynamin II (e, f) was measured using 1.3 Ci [ $\gamma$ - $^{32}$ P]-GTP in the presence or absence of bis-tyrphostin (c, e) or tyrphostin A47 (d, f). The basal activity (open circles) and phospholipid-stimulated activity (solid circles) are compared;

Figure 2: Autoradiograph of nitrocellulose membranes illustrating [ $\alpha$ - $^{32}$ P]-GTP binding to dynamin I and dynamin II is not affected by the addition of bis-tyrphostin or tyrphostin A47 (a, b). Quantitative data is shown in panels (c) and (d);

Figure 3: (a) Graph showing bis-tyrphostin does not act at the PH domain of dynamin I since the compound still inhibits the GTPase activity of a mutant form of recombinant dynamin lacking this domain ("Dynamin I-Delta PH"); (b) photo of an SDS gel stained with Coomassie blue showing that bis-tyrphostin does not block dynamin

binding to lipid, with dynamin being retained in the pellet (P) rather than the supernatant (S);

Figure 4: Fluorimetric assays of exocytosis (a,c) and endocytosis (b,d) in isolated nerve terminals (synaptosomes) shows bis-tyrphostin but not A47 specifically decreases endocytosis. Retrieval efficiency is a more accurate measurement of endocytosis in relation to the preceding amount of exocytosis, and bis-tyrphostin produced a significant block in retrieval efficiency (e).

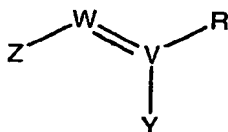
Figure 5: Electron micrographs of isolated rat brain nerve terminals (synaptosomes) showing synaptic vesicle depletion in synaptosomes upon addition of bis-tyrphostin followed by stimulation by depolarisation (a, b) and the accumulation of vesicle invaginations and collared pits (c-h); and

Figure 6: Photographs showing that internalisation of texas-red labelled transferrin into Swiss 3T3 cells (a - d) or HER14 cells (e - h) is inhibited by a 15 minute preincubation with 100  $\mu$ M bis-tyrphostin. DAPI (blue) staining indicates the cell nuclei.

#### Detailed Description of Preferred Embodiments

The carbocyclic group formed when W, V and Y are cyclised in a compound of formula I or III may be a cyclic 5 or 6 membered group which may or may not include any double bonds. Preferably, the carbocyclic group will include one or more double bonds.

Preferably, M and M' of a compound of formula I or III will each independently be a moiety of formula V



Formula V

wherein:

V is C;

W is CH;

Y is hydrogen, cyano, nitro, amino, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, or an unsubstituted C<sub>1</sub>-C<sub>2</sub> group or C<sub>1</sub>-C<sub>2</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; or

- 5 W, V and Y form a 5 or 6 membered substituted or unsubstituted heterocyclic or carbocyclic ring fused with Z, wherein the heterocyclic ring includes from 1 to 3 heteroatoms selected from O, N and S, and the carbocyclic or heterocyclic ring, when substituted, has at least one substituent selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur, or an unsubstituted C<sub>1</sub>-C<sub>2</sub> group or
- 10 C<sub>1</sub>-C<sub>2</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; and

R is CH<sub>2</sub>R', CXR' or CHX'R';

X is O or S;

- X' is cyano, nitro, amino, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy, or an
- 15 unsubstituted C<sub>1</sub>-C<sub>2</sub> group or C<sub>1</sub>-C<sub>2</sub> group substituted with at least one group independently selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; and

R' is NH, O or S bonded to the spacer; and

Z is a group as in formula II.

- 20 Preferably, when W, V and Y are not cyclised, Y will be cyano, nitro, amino, hydroxy, carboxy or thiocarboxy. Most preferably, Y will be cyano.

Preferably, R is CXR' wherein X is O or S and R' is NH, O or S. More preferably, X will be O or S and R' will be NH.

- 25 When Z is a carbocyclic group, the carbocyclic group may be a single cyclic ring or a polycyclic group. The cyclic or polycyclic group may include one or more double bonds. Alternatively, the carbocyclic group may be an aryl group such as benzene or naphthalene, or a polyphenyl group such as bi-phenyl. When the carbocyclic group comprises two rings, the ring bonded directly to W will preferably bear all the substituents, or have at least two substituents when W is CH or linker group or have the, or at least one of, the
- 30 substituents when W, V and Y are cyclised. Preferably, Z will be a group selected from:

- (i) a heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S;
- (ii) a heterocyclic group consisting of one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O, N and S, wherein the heterocyclic group has one or more substituents independently selected from nitro, NH, halo, cyano, amino, hydroxy, carboxy, oxo, sulfur, and C<sub>1</sub>-C<sub>2</sub> alkoxy; and
- (iii) an aryl group consisting of one or two rings independently having 5 or 6 ring members, and at least two substituents independently selected from nitro, NH, amino, halo, cyano, hydroxy, carboxy, oxo, sulfur and C<sub>1</sub>-C<sub>2</sub> alkoxy.

Preferably, when the Z group is a carbocyclic group and has a halo, cyano, C<sub>1</sub>-C<sub>2</sub> alkoxy or C<sub>1</sub>-C<sub>2</sub> acyl substituent, the Z group will also generally have at least two other substituents, preferably independently selected from nitro, NH, amino, hydroxy, carboxy, oxo and sulfur, and most preferably from nitro, NH, amino, hydroxy and carboxy.

Preferably, Z will be a heterocyclic group having one or two rings independently having 5 or 6 ring members including up to 3 heteroatoms selected from O and N, wherein the heterocyclic group has one or more substituents independently selected from nitro, NH, amino, halo, hydroxy, carboxy and oxo, or an aryl group having a single ring of 5 or 6 ring members and at least two substituents independently selected from nitro, amino, halo, hydroxy and carboxy. Preferably, the aryl group will be benzene.

Preferably, when Z is a heterocyclic group, the group will be a substituted or unsubstituted imadazolyl, pyranyl, isobenzylfuranyl, furyl, chromenyl, pyrrolyl,  $\Delta$ H-pyrrolyl, pyrazolyl, pyridyl, pyrazinyl, pyrimidinyl, pyridazinyl, indoliziny, isoindolyl,  $\Delta$ H-indolyl, indolyl, indazolyl, purinyl, quinoliziny, isoquinolyl, quinolyl, pthalazinyl, naphthyridinyl quinoxaliny, quinazolinyl, cinnolinyl, pteridinyl, thienyl, or benzothienyl group. Most preferably, the heterocyclic group will be a substituted group.

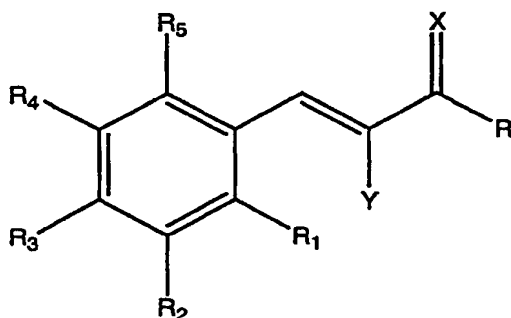
In the instance W, V and Y form a 5 or 6 membered heterocyclic ring fused with Z, the resulting group incorporating Z will typically be a substituted or unsubstituted two ring heterocyclic group. The resulting group may for instance be a substituted or unsubstituted heterocyclic group selected from imadazolyl, chromenyl, indoliziny, isoindolyl, indolyl, indazolyl, purinyl, quinoliziny, isoquinolyl, quinolyl, pthalazinyl,



naphthyridinyl, quinoxalinyl, quinazolinyl, cinnolinyl, pteridinyl, benzothienyl and isobenzofuranyl. Preferably, the resulting group will be substituted or unsubstituted chromenyl, indolyl, or isoquinoline. Again, the heterocyclic group formed by the cyclisation of W, V and Y will preferably be a substituted group.

- 5 Most preferably, a compound of the invention or administered to a mammal in accordance with a method of the invention will be a compound wherein:

M and M' are each independently a moiety of formula VI and are the same or different, and



Formula VI

- 10 X is O or S;

Y is cyano, nitro, amino, halo, hydroxy, sulfhydryl, carboxy, or thiocarboxy;

or

- 15 R<sub>1</sub> and Y are cyclised forming a 5 or 6 membered substituted or unsubstituted heterocyclic or carbocyclic ring, wherein the heterocyclic ring includes 1 or 2 heteroatoms selected from O, N and S, and the carbocyclic or heterocyclic ring, when substituted, has at least one substituent selected from cyano, nitro, NH, amino, oxo, halo, hydroxy, sulfhydryl, carboxy, thiocarboxy and sulfur; and

- 20 R<sub>2</sub> to R<sub>5</sub> are independently hydrogen or a substituent independently selected from nitro, amino, halo, hydroxy, carboxy, sulfhydryl, thiocarboxy, C<sub>1</sub>-C<sub>2</sub> alkoxy and C<sub>1</sub>-C<sub>2</sub> acyl; or

$R_1$  to  $R_5$  are independently hydrogen or a substituent independently selected from nitro, amino, halo, hydroxy, carboxy, sulfhydryl, thiocarboxy, halo,  $C_1$ - $C_2$  alkoxy and  $C_1$ - $C_2$  acyl; and

R is NH, O is S bonded to the spacer Sp;

5 wherein for at least for one of M and M', at least two of  $R_1$  to  $R_5$  are other than hydrogen and when  $R_1$  to  $R_2$  are other than hydrogen at least one of  $R_3$  to  $R_5$  is also other than hydrogen, or when  $R_1$  and Y are cyclised, at least two of  $R_2$  to  $R_5$  are other than hydrogen when Y and  $R_1$  form an unsubstituted carbocyclic group or at least one of  $R_2$  to  $R_5$  is other than hydrogen when Y and  $R_1$  form a heterocyclic  
10 group.

Preferably, when Y and  $R_1$  are not cyclised and  $R_1$  and  $R_2$  are other than hydrogen,  $R_3$  will also be other than hydrogen. Typically, the at least two substituents of  $R_1$  to  $R_5$  will be in an ortho position relative to one another. When the compound has three substituents it is preferred the substituents are adjacent to each other. Preferably, in this instance, either  
15  $R_1$  to  $R_3$  are other than hydrogen or  $R_2$  to  $R_5$  are other than hydrogen.

Typically, when at least one of  $R_1$  to  $R_5$  or  $R_2$  to  $R_5$  is halo,  $C_1$ - $C_2$  alkoxy or  $C_1$ - $C_2$  acyl, there will be at least one other substituent selected from nitro, amino, hydroxy, carboxy or thiocarboxy when  $R_1$  and Y are cyclised and form a heterocyclic ring, or at least two other substituents selected from those substituents when  $R_1$  and Y are not cyclised or form an  
20 unsubstituted carbocyclic ring.

Halo substituents will typically be selected from fluoro, chloro, bromo, and iodo. Preferably, a halo substituent will be selected from fluoro and chloro.

Preferably, the spacer moiety Sp of a compound of the invention or administered to a mammal in accordance with the invention will have a hairpin conformation and  
25 preferably, will be a substituted or unsubstituted 1 to 7 atom chain which may include one or more atoms other than carbon such as N, O or S, and one or more double bonds. The spacer may be substituted with one or more groups independently selected from hydroxy, amino, halo and nitro, or other group which does not substantially affect the flexibility or conformation of the chain. Most preferably, the spacer moiety will be a substituted or  
30 unsubstituted alkane chain. Typically, the spacer will be unsubstituted an alkane chain having the structure:



wherein n is an integer of from 1 to 5 and more usually 2 or 3.

Endocytosis is a major contributor or direct cause of diverse human diseases. A list of vesicle trafficking-specific diseases has been published, see for example Aridor and Hannan 2000, Traffic 1:836-851 and Aridor and Hannan 2002, 3:781-790 the contents of which are incorporated herein by reference in their entirety. Accordingly, methods of the invention may for instance be useful in the prophylaxis or treatment of cancers, ophthalmologic disease, immunodeficiency diseases, gastrointestinal diseases, viral and bacterial infections, other pathogenic infections, neurodegenerative, neurological and kidney diseases and conditions, and other disorders which involve dynamin-dependent endocytosis, or which are otherwise sensitive to inhibition of dynamin-dependent endocytosis.

For example, it is known that human polyomavirus JCV is the etiologic agent of progressive multifocal leukoencephalopathy, a fatal central nervous system (CNS) demyelinating disease and its entry to neurons is blocked by endocytosis inhibitors such as chlorpromazine (Atwood W., 2001). Similarly, infection by HIV (Wyss S. *et al.*, 2001), influenza virus (Roy A., *et al.* 2000) and adeno-associated virus (Duan D. *et al.*, 1999) is by endocytosis or is sensitive to its inhibitors.

In addition, growth factor receptors (e.g. EGF-R) require dynamin for internalisation and maintenance of cellular activities from signalling to cell growth (Seto E. *et al.*, 2002). Blocking endocytosis with dynamin constructs prevents cell proliferation in many of these examples (Grieb T. *et al.*, 2000) and provides evidence that dynamin II (the non-neuronal form) inhibitors may have anti-cancer activity. Dent's disease (polycystic kidney disease) also involves endocytosis of ClC-5 chloride channel and endocytosis blockers prevent its internalisation (Schwake M. *et al.*, 2001).

Dynamin is central to all endocytic trafficking from the cell surface, the Golgi apparatus, endosomes and mitochondria. Several neurodegenerative diseases are associated with these trafficking pathways. Two are implicated in generation of  $\beta$ -amyloid, namely the endocytic and the secretory pathways (Aridor & Hannan 2000). In the brain, disease and conditions in which endocytosis plays a role include Alzheimer's disease, Huntington's disease (HD), stiff-person syndrome, Lewy body dementias, and Niemann-

Pick type C disease (Cateldo et al., 2001; Metzler et al., 2001; Ong et al., 2001; Smith et al., 2000).

In Alzheimer's disease  $\beta$ -amyloid precursor protein (APP) is internalized from axonal cell surfaces in clathrin-coated vesicles and sorted away from recycling synaptic vesicles, and transported to endosomes and the cell soma (Marquez-Sterling N. *et al.*, 1997). The endosome is the first compartment along the dynamin-dependent endocytic pathway after internalization of APP or ApoE (Smythies J., 2000) and endosomal alterations are evident in pyramidal neurons in Alzheimer brain (Cataldo A. *et al.*, 1997). Endocytic pathway activation is prominent in APP processing and -amyloid formation and is an early feature of neurons in vulnerable regions of the brain in sporadic Alzheimer's disease (Cataldo A. *et al.*, 2001).

Huntington's disease (HD) is a neurodegenerative disorder principally affecting striatal neurons, yet the mutated gene product huntingtin is not brain-specific. Huntingtin interacts strongly with members of the Huntingtin-interacting protein 1 (HIP1) family. The huntingtin-HIP1 interaction is restricted to the brain and is inversely correlated to the polyglutamine length in huntingtin. Loss of normal huntingtin-HIP1 interaction may contribute to a defect in membrane-cytoskeletal integrity in the brain. HIP1 is a fundamental component of the dynamin-mediated endocytic machinery (Metzler M. *et al.*, 2001). Hence, numerous reports have linked the neurological defects in HD to endocytosis abnormalities (Aridor & Hannan, 2000; Metzler M. *et al.*, 2001).

Another example is the presynaptic synuclein protein which is a prime candidate for contributing to Lewy body diseases, including Parkinson's disease, Lewy body dementia and a Lewy body variant of AD. Exogenous synuclein causes neuronal cell death due to its endocytosis and formation of intracytoplasmic inclusions. Cell death and  $\alpha$ -synuclein aggregates are direct consequences of its endocytosis in human neuroblastoma cells (Sung J. *et al.*, 2001).

Endocytosis has also been implicated in epilepsy. For example, mice with targeted disruption of either of two endocytic proteins synaptojanin (SJ) or amphiphysin have reduced SVE and die from random seizures throughout their lives (Di Paolo et al., 2002) indicating a role in neuronal excitability and a link to epilepsy.

Endocytic pathways are also utilized by viruses, toxins and symbiotic microorganisms to gain entry into cells. For instance, botulism neurotoxins and tetanus neurotoxin are bacterial proteins that inhibit transmitter release at distinct synapses and cause two severe neuromuscular diseases, tetanus and botulism. Their action is dependent on their internalisation via endocytosis into nerve terminals (Humeau et al., 2000). Hence targeting endocytosis with inhibitors has application as a clinically useful strategy.

Accordingly, examples of specific diseases and conditions for which methods of the invention may be useful for the prophylaxis or treatment of include but are not limited to, multifocal leukoencephalopathy, polycystic kidney disease,  $\beta$ -amyloid associated diseases, Alzheimer's disease, Huntington's disease, stiff-person syndrome, Lewy body diseases, Lewy body dementias, Parkinson's disease, epilepsy, tetanus, botulism, HIV infection, influenza and mucopolysaccharidosis.

Preferably, the compound of formula I administered to a mammal in accordance with the invention will be a dimeric benzylidenemalonitrile tyrphostin. Most preferably, the dimeric tyrphostin will be bis-tyrphostin or an analogue thereof. With knowledge of the features and/or groups of bis-tyrphostin or dimeric tyrphostin that provide the ability to bind to and inhibit the activity of dynamin, analogues and more particularly mimetics may be designed that while differing in structure nevertheless retain this capacity. The use of dimeric tyrphostin analogues and particularly analogues of bis-tyrphostin in methods described herein is expressly encompassed by the present invention.

The term "analogue" encompasses a molecule that differs from the original molecule but retains similarity in one or more features that provide the biological function or activity characteristic of the original molecule. An analogue may have substantial overall structural similarity with the original molecule or only structural similarity with one or more regions of the original molecule responsible for the provision of the biological function or activity, or which otherwise have involvement in the provision of the biological function or activity. An analogue of bis-tyrphostin may for instance be provided by substituting one or both hydroxy substituents on one or both aromatic groups of the compound with another suitable group or a number of different suitable groups as described above. Alternatively, or as well, one or more other groups of the compound may be removed, modified or replaced.

The design of an analogue may involve determining the physical properties of the original compound such as size, charge distribution and tertiary structure and identifying which features of the compound are necessary for retaining the capacity to bind to dynamin. In particular, the original compound may be modelled taking into account the stereochemistry and physical properties of the compound utilising x-ray crystallography, nuclear magnetic resonance and commercially available computer modelling software. In a preferred variation of this approach, the modelling will take into account the interaction of the compound with dynamin itself such that any change in conformation arising from the interaction may be considered in the design of the analogue. Such modelling techniques are well known in the art and are well within the scope of the skilled addressee.

The provision of an analogue added can also involve selecting or deriving a template molecule onto which chemical groups are added to provide the required physical and chemical characteristics, or for facilitating further chemical reactions for obtaining the required physical and chemical characteristics. The selection of template molecule and chemical groups is based on ease of synthesis, risk of potential for degradation *in vivo*, stability and maintenance of biological activity upon administration. Pharmacological acceptability and the like are also taken into consideration in the design as is understood by the skilled addressee.

Compounds may be administered in accordance with the invention with one or more other compounds or drugs. For example, a compound may be co-administered to the subject mammal in combination or in conjunction with chemotherapeutic drugs or drugs conventionally used in the prophylaxis or therapeutic treatment of the particular disease or condition for which the mammal is being treated. By "co-administered" is meant simultaneous administration in the same formulation or in two different formulations by the same or different routes, or sequential administration by the same or different routes. By "sequential" administration is meant administration one after the other which may involve a time delay between administration of the compound and the other drug or drugs ranging from very short periods up to hours or days.

Suitable pharmaceutical compositions include solutions suitable for injection. Such injectable compositions will be fluid to the extent that syringability exists and typically, will be stable for at least several months to allow for storage after manufacture. The carrier may be a solvent or dispersion medium containing one or more of surfactants,

physiological saline, ethanol, polyol, (e.g. glycerol, propylene glycol, liquid polyethylene glycol and the like), vegetable oils, and mixtures thereof.

For oral administration, the compound may be formulated with an orally acceptable inert diluent, an assimilable edible carrier or it may for instance, be enclosed in a hard or soft shell gelatin capsule. Alternatively, it may be added directly to food. Moreover, the compound may be incorporated with one or more excipients such as dicalcium phosphate, a disintegrating agent such as corn starch, potato starch, or alginic acid and used in the form of ingestible tablets, buccal tablets, troches, capsules, elixirs, suspensions and syrups.

Tablets, pills and the like may also contain one or more of a binder such as gum tragacanth, acacia, corn starch or gelatin, a lubricant such as magnesium stearate, a sweetening agent such as sucrose, lactose, saccharin, and a flavouring agent. When the dosage form is a capsule, it may contain a liquid carrier in addition to one or more of the above ingredients. Various other ingredients may be present as coatings. In addition, the compound may be incorporated into any suitable sustained release preparation or formulation.

The compound will typically be formulated into a pharmaceutical composition with a pharmaceutically acceptable carrier or excipient for administration to the intended subject. Any conventionally known such carriers diluents and excipients deemed suitable may be used. Suitable pharmaceutically acceptable carriers and excipients include any known appropriate solvents, dispersion media and isotonic preparations or solutions. Use of such ingredients and media for pharmaceutically active substances is well known. Typically, a composition of the invention will also incorporate one or more preservatives such as parabens, chlorobutanol, phenol, sorbic acid, and thimersal.

It is particularly preferred to formulate parenteral compositions in dosage unit form for ease of administration and uniformity of dosage. Dosage unit form as used herein is to be taken to mean physically discreet units suited as unitary dosages for the subject to be treated, each unit containing a predetermined quantity of active agent calculated to produce the desired prophylactic or therapeutic effect in association with the carrier and/or excipient selected.

The dosage of the compound to be administered will depend on a number of factors including whether the compound is to be administered for prophylactic or therapeutic use, the condition for which the agent is intended to be administered, the severity of the condition, the age of the subject, and related factors such as weight and general health of the subject as may be determined by the physician or medical attendant in accordance with accepted principles. For example, a low dosage may initially be given which is subsequently increased following evaluation of the subject's response. Similarly, frequency of administration may be determined in the same way that is, by continuously monitoring the subject's response between each dosage and if necessary, increasing the frequency of administration or alternatively, reducing the frequency of administration.

The route of administration of a pharmaceutical composition will again depend on the nature of the disease or condition for which the composition is to be administered. Suitable routes of administration may include but are not limited to respiratoraly, intratrachealy, nasopharyngealy, intravenously, intraperitonealy, subcutaneously, intraderamaly, intramuscularly, by infusion, orally, rectally, topically and by slow-release implant. In the case of intravenous routes, particularly suitable routes are via injection into blood vessels which supply a tumour or particular organs to be treated. Compounds may also be delivered into cavities such as for example the pleural or peritoneal cavity, or be injected directly into tumour or afflicted tissue.

The invention will now be described hereinafter with reference to a number of non-limiting examples.

#### **Example 1: Identification of tyrphostin inhibitors of dynamin GTPase activity**

##### **1.1 *Materials and assays***

Phosphatidylserine, 1,2-diolein, calmodulin, ATP, GTP, leupeptin, phenylmethylsulfonylfluoride, Tween 80, bis(sulfosuccinimidyl) suberate (BS3) and glutathione agarose were obtained from Sigma. Papain and antipain-dihydrochloride were obtained from Boehringer Mannheim (Federal Republic of Germany). Gel electrophoresis reagents and equipment were sourced from Bio-Rad. [ $\gamma$ - $^{32}$ P]ATP (3000 Ci/mmol) and [ $\gamma$ - $^{32}$ P]GTP (25  $\mu$ Ci/mmol) were from Amersham plc, UK. Protein molecular weight markers and chromatography resins were sourced from Pharmacia. All other reagents were of analytical reagent grade or better.



### 1.1.1 *Production of proteins*

The plasmid for GST-Amph2-SH3 (muscle Amph2) (Butler et al., 1997) was provided by Pieter DeCamilli, Yale, Connecticut, USA, in pGEX2T vectors. The plasmid was grown in *E. coli* and the GST-Amph2-SH3 fusion protein was purified on glutathione (GSH)-Sepharose by elution with 10 mM reduced GSH in 20 mM Tris-HCl, pH 7.5, dialysed against the same buffer without GSH and stored at 4°C. Dynamin was purified from sheep brain by extraction from the peripheral membrane fraction of whole brain (Robinson et al., 1993) and affinity purification on GST-Amph2-SH3-sepharose as previously described (Marks and McMahon, 1998), yielding 8 mg protein from 250 g sheep brain. Recombinant dynamin II was expressed in insect cells and was a gift from Dr Sandra Schmid (Scripps, San Diego, CA). Recombinant dynamin I lacking the PH domain (dynamin PH, provided by Robin Scaife) was expressed in insect cells using baculoviral infection (Salim et al., 1996).

### 1.1.2 *GTPase assay*

Dynamin GTPase activity was determined by hydrolysis of [ $\gamma$ -<sup>32</sup>P]GTP by a method modified from that described previously (Robinson et al., 1993). Briefly, purified dynamin I or dynamin II (0.2 µg/tube) was incubated in GTPase buffer (10 mM Tris, 10 mM NaCl, 2 mM Mg<sup>2+</sup>, 0.05% Tween 80, pH 7.4, 1 µg/ml leupeptin and 0.1 mM PMSF) and a GTP cocktail containing 0.3 mM GTP and 1.3 µCi [ $\gamma$ -<sup>32</sup>P]-GTP in the presence or absence of varying concentrations of inhibitors or DMSO vehicle for 10 min at 30°C. The final assay volume was 40 µl. Dynamin activity was measured as either basal or phospholipid-stimulated with the addition of 5 µg/ml L-phosphatidylserine. The reaction was terminated with 100 µl of GTPase stop buffer (2% formic acid, 8% acetic acid, pH 1.9), followed by 600 µl of acid-washed charcoal solution (7% charcoal in acidic solution (w/v)) and 100 µl BSA (5 mg/ml). After centrifuging for 5 min (13,000 rpm at room temperature), 200 µl of each supernatant was counted in a  $\beta$ -counter for the release of <sup>32</sup>P<sub>i</sub> from [ $\gamma$ -<sup>32</sup>P]-GTP.

### 1.1.3 [ $\alpha$ -<sup>32</sup>P]-GTP binding assay

The [ $\alpha$ -<sup>32</sup>P]-GTP-binding assay was performed in the wells of a 96-well microtitre plate. Dynamin (0.2 µg/well) was added to GTPase buffer and incubated for 10 min at 4°C in the dark. [ $\alpha$ -<sup>32</sup>P]-GTP (2 µCi/tube) was then added to the reaction and incubated for a

further 10 min at 4 °C in the dark. The microtitre plate was then irradiated with a short wavelength ultraviolet lamp at 315 nm for 30 min at a distance of 8 cm. The specificity of photolabelling was determined by comparing the labelling in the presence and absence of 1 mM cold GTP. Samples were then applied to nitrocellulose membranes by aspiration  
 5 through the wells of a 24 well slot blotter. The nitrocellulose was washed 3 times with PBS and dried. Bound nucleotide was detected by a phosphorimager (Molecular Dynamics).

#### 1.1.4 *Phospholipid binding and helix assembly*

Dynamin I (50 µg/ml) purified from whole sheep brain was incubated with phosphatidylserine liposomes (80 µg/ml, sonicated into 30 mM Tris/HCl pH 7.4) in 100 µl  
 10 of assembly buffer (1 mM EGTA, 30 mM Tris, 100 mM NaCl, 1 mM DTT, 1 mM PMSF, and Complete protease inhibitor cocktail tablet (Roche)) in the presence or absence of 1 mM Mg/GTP for 1 hour at 25°C. The samples were centrifuged at 14,000 rpm for 15 min to separate lipid-bound (P) and free (S) dynamin and the fractions analysed by gel  
 15 electrophoresis on a 12 % SDS polyacrylamide gel. When present, drugs (10 µM and 100 µM) were pre-mixed with the phospholipid before incubating with dynamin I.

#### 1.1.5 *Texas red-transferrin uptake in cells*

Transferrin (Tf) uptake was analysed in Swiss 3T3 and HER14 cells based on methods previously described (van der Bliek et al., 1993). Briefly, cells were plated to 60%  
 20 confluency in DMEM medium plus 10% foetal calf serum after which the cells were incubated overnight (8-10 hours) in DMEM minus foetal calf serum. Texas red-transferrin (Tf-TxR, Molecular Probes, Oregon) was added to a final concentration of 5 µg/ml and the cells incubated at 37°C for 10 minutes. Cell surface staining was removed by incubating  
 25 the cells in an ice cold acid wash solution (0.2 M acetic acid + 0.5 M NaCl, pH 2.8) for 15 minutes. Cells were immediately fixed with 4% paraformaldehyde for 10 minutes then washed 3 times with PBS. Nuclei were stained using DAPI (Molecular Probes, Oregon). Slides were mounted using DABCO and the fluorescence was monitored using a Leica DMLB bright field microscope and SPOT digital camera. In experiments with inhibitors, the DMEM was supplemented with bis-tyrphostin 15 or 60 minutes before the addition of Tf-TxR.

### 1.1.6 *Endocytosis*

Isolated nerve terminals (synaptosomes) were prepared from rat cerebral cortex by centrifugation on discontinuous percoll gradients (Dunkley et al., 1986). Fractions 3 and 4 were pooled and used in all experiments. Endocytosis was measured using uptake of the fluorescent dye FM2-10 as previously described (Cousin and Robinson 2000a).

Synaptosomes (0.6 mg in 2 ml) were incubated for 5 min at 37°C in plus or minus  $\text{Ca}^{2+}$  Krebs-like solution. FM2-10 (100 M) was added 1 min before stimulation with 30 mM KCl (S1). As FM2-10 is taken up by vesicles via endocytosis at the S1 phase of stimulation, synaptosomes were incubated with antagonists during this phase. Specifically, synaptosomes were incubated with tyrphostin A47 or bis-tyrphostin for 5 min prior to stimulation. After 2 min of stimulation synaptosomes were washed twice in plus  $\text{Ca}^{2+}$  solution containing 1 mg/ml bovine serum albumin. The washing steps remove non-internalised FM2-10 and the tyrphostins. Washed synaptosomes were resuspended in plus  $\text{Ca}^{2+}$  solution at 37°C, transferred to a fluorimeter cuvette and stimulated with a standard addition of 30 mM KCl (S2). The standard S2 stimulation releases all accumulated FM2-10 and allows endocytosis to be measured as the decrease in FM2-10 fluorescence due to dye release into solution (excitation 488 nm, emission 540 nm).

Endocytosis was calculated as the decrease in absolute fluorescence stimulated by 30 mM KCl at S2. The displayed traces represent the average release of FM2-10 from synaptosomes after subtraction of background traces acquired from synaptosomes loaded with FM2-10 in the absence of  $\text{Ca}^{2+}$ . Retrieval efficiency is a more accurate measure of endocytosis since it takes into account the amount of prior exocytosis. Retrieval efficiency was calculated as endocytosis/exocytosis, where endocytosis is defined as above and exocytosis as  $\text{Ca}^{2+}$ -dependent glutamate release after 2 min of stimulation. The retrieval efficiency value was normalised to a ratio of 1.0 for 30 mM KCl.

### 1.1.7 *Glutamate release assay*

The glutamate release assay was performed using enzyme-linked fluorescent detection of released glutamate (Cousin and Robinson., 2000a, b). Briefly, synaptosomes (0.6 mg in 2 ml) were resuspended in either plus (1.2 mM  $\text{CaCl}_2$ ) or minus (1 mM EGTA)  $\text{Ca}^{2+}$  Krebs-like solution (118.5 mM NaCl, 4.7 mM KCl, 1.18 mM  $\text{MgCl}_2$ , 0.1 mM  $\text{Na}_2\text{HPO}_4$ , 20 mM Hepes, 10 mM glucose, pH 7.4) at 37°C. Experiments were started after addition of 1 mM NADP<sup>+</sup>. After 1 minute 50 U of glutamate dehydrogenase was added and the

synaptosome suspension was stimulated after 4 minutes with 30 mM KCl. Increases in fluorescence due to production of NADPH were monitored in a Perkin-Elmer LS-50B spectrofluorimeter at 340 nm excitation and 460 nm emission. Experiments were standardised by the addition of 4 nmol of glutamate. Data is presented as  $\text{Ca}^{2+}$ -dependent glutamate release, calculated as the difference between release in plus and minus  $\text{Ca}^{2+}$  solution for identical stimulation conditions. In experiments using inhibitors, synaptosomes were preincubated for 5 min with either tyrphostin A47 or bis-tyrphostin before stimulation with KCl.

#### 1.1.8 *Electron microscopy*

Synaptosomes were incubated for 5 min in Krebs-like solution containing 1.2 mM  $\text{Ca}^{2+}$  then stimulated with 30 mM KCl for 2 minutes. Synaptosomes were preincubated with 100  $\mu\text{M}$  bis-tyrphostin 5 minutes prior to KCl addition where indicated. After stimulation, synaptosomes were pelleted in a microfuge for 1 minute at room temperature then fixed by gentle resuspension in ice-cold phosphate buffered saline supplemented with 5% glutaraldehyde. After 1 hr they were centrifuged at low speed (2500 rpm) for 5 min at room temperature to loosely pellet the synaptosomes. The pellets were washed gently 3 times with MOPs buffer with low spins (2500 rpm) for 7 minutes then gently resuspended in a 10% bovine serum albumin (BSA) in water and allowed to stand for 20 min at room temperature. The synaptosomes were then centrifuged again for 7 minutes at low speed (2500 rpm), overlaid with Karnovsky's fixative and incubated at 4°C overnight. The pellets were subsequently rinsed and fixed in a buffered solution of osmium tetroxide for 3 hours. Synaptosomes were then rinsed and stained for 1 hour in 2% aqueous uranyl acetate prior to being dried by a series of sequential 10 minute washes: 50% ethanol plus 0.1% NaCl, 70% ethanol plus 0.1% NaCl, 95% ethanol plus 0.1% NaCl, 100% ethanol plus 0.1% NaCl twice and 100 % acetone twice. They were then infiltrated with an acetone/resin mixture (1:1) for 1 hour, washed 3 times for 10 minutes in Spur's epoxy resin at 70°C, then embedded within flat molds filled with Spur's epoxy resin for 10 hours at 70°C.

An ultramicrotome Ultracut-E (Reichert, Germany) was used to obtain 0.5  $\mu\text{m}$  epoxy sections from the resin blocks. The sections were cut with a diamond knife (Diatome, Switzerland), floated on water drops, placed on electron microscopy grids and double stained: first using 2% uranyl acetate in ethanol for 15 minutes and then Reynold's lead citrate for 4 minutes. The grids were washed in water, touch dried using absorbent

filter paper and stored until analysis with an electron microscope. Analyses were performed on a Phillips 1L-BioTwin (Eindhoven, Netherlands) electron microscope and pictures taken were printed on electron microscope plate film (Kodak, 4489, 8.3 cm X 10.2 cm).

## 5 1.2 *Results*

### 1.2.1 Bis-tyrphostin inhibits the GTPase activity of both dynamin I and dynamin II

The GTPase activity of dynamin plays an essential role in the ability of vesicles to bud from the plasma membrane during endocytosis. To initially find an inhibitor of the neuron-specific dynamin I a number of protein kinase inhibitors and some lipid kinase inhibitors which are highly potent ATPase active site-directed inhibitors were tested. These compounds were selected on the basis of the hypothesis that as ATPase active sites are similar to GTP active sites, then some ATPase inhibitors may also target dynamin. The results obtained showed some success with low potency inhibition of dynamin I GTPase activity.

15 A series of tyrphostins were then evaluated and two were found that showed inhibition, namely tyrphostin A47 ( $IC_{50} = 100 \mu M$ ) and the most potent inhibitor of this sampling bis-tyrphostin, which showed an  $IC_{50}$  of  $2 \mu M$  (see Fig. 1c and 1d).

20 Tyrphostin A47 and bis-tyrphostin were subsequently tested for the purpose of evaluating whether the observed inhibition was specific to dynamin I, or if it also affected the ubiquitous dynamin II. Both drugs proved to be more potent for dynamin II (see Fig. 1e and 1f). More particularly, tyrphostin A47 showed an  $IC_{50}$  of  $9 \mu M$  for dynamin II while bis-tyrphostin showed an  $IC_{50}$  of just  $0.5 \mu M$ . This indicates that a drug specific to each dynamin gene product may be designed thereby allowing for the pharmaceutical control of various forms of endocytosis.

### 25 1.2.2 *Bis-tyrphostin and tyrphostin A47 do not prevent GTP binding to dynamin I or dynamin II*

To evaluate the mechanism of action of bis-tyrphostin and tyrphostin A47 in preventing GTP hydrolysis by dynamin, the drugs were tested to see if they were competing with GTP at the active site on dynamin. [ $\alpha$ - $^{32}P$ ]-GTP binding assays were completed to visualise radiolabelled GTP binding to dynamin in the presence or absence of

bis-tyrphostin, tyrphostin A47 or BIM I (Fig. 2a-d). The controls (no drug) showed that  $[\alpha\text{-}^{32}\text{P}]\text{-GTP}$  did bind to dynamin. In the presence of bis-tyrphostin and tyrphostin A47 GTP binding was not seen to decrease but, at high concentrations, was seen to actually be enhanced. This is especially so in the case of tyrphostin A47 vastly increasing GTP binding to dynamin II at high concentrations. The GTPase inhibitor BIM I was also found to compete with GTP for binding to dynamin as seen by the decrease in  $[\alpha\text{-}^{32}\text{P}]\text{-GTP}$  binding.

Competition of these drugs with GTP for the active site of 4 small G proteins (Rab3A, Ras, Arf2, RalA) was also tested. It was found that neither drug affected  $[\alpha\text{-}^{32}\text{P}]\text{-GTP}$  binding to these proteins (data not shown). This indicates that bis-tyrphostin and tyrphostin A47 are likely to be specific in their action to dynamin and not other G proteins which may be present in the nerve terminal or cell.

### 1.2.3 *Bis-tyrphostin does not act at the PH domain of dynamin I*

In order to determine if bis-tyrphostin was inhibiting dynamin I via its PH domain, the effect of bis-tyrphostin on a recombinant version of dynamin I lacking the PH domain (PH domain dynamin I) was compared to its effect on wild type dynamin I (Fig. 3a). The PH domain of dynamin I acts as a negative regulator of its GTPase activity, PH domain dynamin I is constitutively active and not affected by phospholipids. The results show that bis-tyrphostin was still able to inhibit PH domain dynamin I GTP hydrolysis more than 50% at 10  $\mu\text{M}$ . This shows that the PH domain is not the site of action of bis-tyrphostin on dynamin I which means that bis-tyrphostin must be inhibiting at an allosteric site on the dynamin I molecule. BIM I, however, lost its ability to inhibit dynamin I GTPase activity with the removal of the PH domain showing that this drug does prevent GTP hydrolysis via the PH domain.

Dynamin interaction with phospholipids stimulates GTPase activity by inducing cooperative dynamin helix assembly. Assembled dynamin is readily detected by a simple sedimentation assay and this characteristic was used to determine whether bis-tyrphostin regulates dynamin helix assembly or phospholipid interaction. Dynamin alone does not sediment in the assay and is retained in the supernatant (Fig 3b, lanes 1-2), while it is found largely in the pellet in the presence of PS liposomes (lanes 3-4). Phospholipid binding, and hence dynamin helix assembly, was completely unaffected by 10 or 100  $\mu\text{M}$  bis-tyrphostin (lane 5-12).  $\text{Mg/GTP}$  was added to the assay but did not alter the result. This indicates

that bis-tyrphostin does not prevent dynamin association with phospholipids, nor its cooperative assembly. Hence, bis-tyrphostin inhibits dynamin GTPase activity at an allosteric site, and that it inhibits after the helix has assembled.

#### 1.2.4 *Bis-tyrphostin, but not tyrphostin A47, inhibits dynamin I-mediated synaptic vesicle retrieval, forming dynamin I rings in the process*

Fluorimetry was used to determine the effect of bis-tyrphostin and tyrphostin A47 on SVE in a population of rat brain nerve terminals (synaptosomes, Fig.4). Bis-tyrphostin and A47 had no effect on exocytosis ( $\text{Ca}^{2+}$ -dependent glutamate release, Fig 4a and 4b). Bis-tyrphostin (100  $\mu\text{M}$  for 10 min) significantly inhibited SVE, whereas A47 (100  $\mu\text{M}$ ) did not (Fig 4c and d). Since the amount of SVE detected in this assay is dependent on the prior extent of exocytosis, the inhibition of endocytosis was quantified by calculating retrieval efficiency. This parameter is a ratio of the amount of endocytosis divided by the amount of exocytosis for each drug (Cousin et al., 2001). A retrieval efficiency of 1 indicates no drug effect on endocytosis. Tyrphostin A47 produced a retrieval efficiency of 0.95 ( $\pm 0.05$ ) and bis-tyrphostin of 0.7 ( $\pm 0.05$ , Fig 4e). This indicates a significant reduction in SVE by bis-tyrphostin.

Since bis-tyrphostin inhibits dynamin I GTPase activity (Fig 1), but not GTP binding (Fig 2), a study was undertaken to determine whether bis-tyrphostin might also trap dynamin at the specific stage in SVE wherein it assembles as rings around the necks of budding synaptic vesicles. Synaptosomes at rest or depolarised once for 10 sec in 41 mM KCl (S1) exhibited normal morphology by electron microscopy (EM) (Fig 5a-b). Nerve terminals were characterised by: i) a smooth, sealed plasma membrane, ii) they were completely filled with small synaptic vesicles, and iii) they almost always contained one to three normal mitochondrial profiles and occasionally contained a synapse and associated postsynaptic density. When unstimulated synaptosomes were treated with bis-tyrphostin there was no effect on their morphology (Fig 5a). However, when depolarised there was a massive depletion of synaptic vesicles (Fig 5b). A small number of plasma membrane invaginations were also detected (Fig 5e-f), suggestive of failed endocytosis. As predicted for a blocker of GTP hydrolysis but not GTP binding, a number of collared pits were observed, with vesicle necks clearly encircled by dense collars (Fig 5c, d, g and h).

### 1.2.5 *Bis-tyrphostin blocks the dynamin II-mediated receptor-mediated endocytosis of transferrin into Swiss 3T3 cells and HER14 cells*

Transferrin is transported into cells by the process of receptor-mediated endocytosis which is mediated by dynamin II. The effect of both bis-tyrphostin and tyrphostin A47 on  
 5 transferrin internalisation into non-neuronal cells was tested (Figure 6). Control cells showed a large degree of cytoplasmic staining (panels a and e) indicating that transferrin has been internalised into the cells. The cell nuclei were co-stained in blue with DAPI to indicate the location of the cell bodies (panels b, d, f and h). Upon addition of bis-tyrphostin a very large decrease in transferrin staining was observed. Tyrphostin A47 also  
 10 produced this effect though not as dramatically as bis-tyrphostin (not shown). The inhibition was also found to be concentration-dependent. The vehicle DMSO had no effect on transferrin internalisation.

### 1.3 *Discussion*

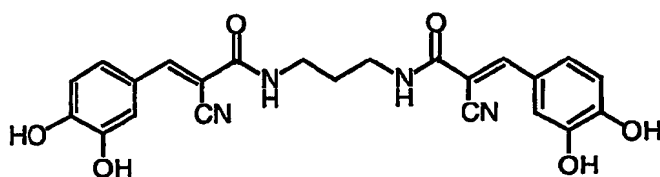
As first demonstrated in the mutant *Drosophila* strain *shibire*, blocking dynamin  
 15 and endocytosis in nerve terminals results in a dramatic depletion of most SVs. Since the large number of SVs are one of the most defining morphological features of nerve terminals their loss is readily evident visually. Furthermore, the resulting morphology of the plasma membrane is known to provide a strong indication of the point in endocytosis at which the block is occurring. Bis-tyrphostin depleted nerve terminals of most SVs and  
 20 produced a very small number of vesicles trapped on the plasma membrane with clear dynamin collars or rings around their necks. This dramatic result revealed that the site of action of bis-tyrphostin follows ring assembly and before neck fission. However, surprisingly, dynamin collars were rare. This surprising complexity suggests bis-tyrphostin blocks at a second point prior to ring assembly providing support that dynamin  
 25 GTPase activity is important at two distinct points in the mechanisms of SVE.

The three dynamin gene products may mediate at least 3 forms of endocytosis. Dynamin I mediates SVE, dynamin II mediates RME and dynamin III may mediate endocytosis in postsynaptic spines (Gray et al., 2003). Further mechanistic subtleties are also known. Differential inhibition of the dynamins provides the capability of  
 30 distinguishing between these cellular roles. In particular, a selective inhibitor is an important tool for discriminating between different types of endocytosis and has clinical interest for targeting pathology based on the different forms of endocytosis.



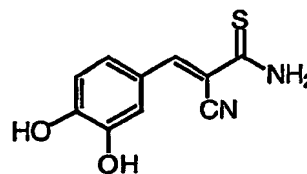
The results further indicate that bis-tyrphostin (BisT or AG537) inhibits the GTPase activity of dynamins I and II and blocks both SVE in nerve terminals (synaptosomes) and RME of transferrin in 3T3 or HER14 cells. Its site of action is not the GTP binding site nor the PH domain and so it is an allosteric inhibitor. Since it does not affect GTP binding it also should not affect dynamin assembly into rings. This provides a unique tool that targets dynamin after it has assembled. Bis-tyrphostin has previously been found to inhibit EGFR-TK ( $IC_{50} = 0.4 \text{ M}$ ) and EGF-dependent cell proliferation ( $IC_{50} = 3 \text{ M}$ ) (Gazit et al., 1996). Therefore, analogues were designed that retained dynamin inhibition, but which lose their effect on EGFR-tyrosine kinase (since the determinants for tyrosine kinase specificity are well known (Gazit et al., 1996)).

#### Example 2: Development of tyrphostin analogues



1

$IC_{50} = 2 \text{ } \mu\text{M}$



2

$IC_{50} = 70 \text{ } \mu\text{M}$

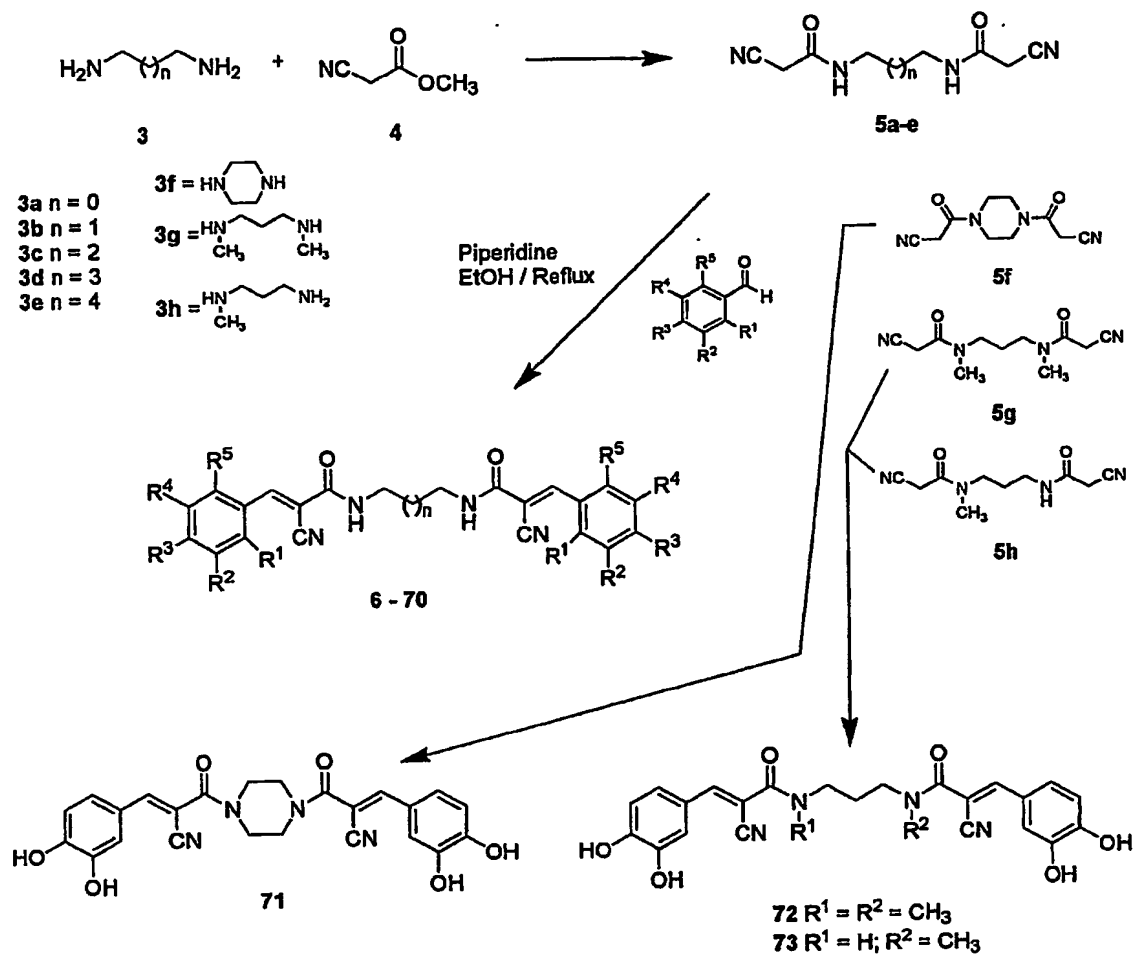
Structures of bis-tyrphostin (1) and tyrphostin A47 (2).

#### 2.1 Development of analogues

The structure for bis-tyrphostin and tyrphostin A47 are shown above. The structural similarities between these compounds of the 3,4-dihydroxybenzene and the presence of the cyanoamide or thioamide suggested that these groups may be important for dynamin inhibition. These features are highly amenable to solution phase parallel synthesis approaches to library development and two libraries were synthesised to determine type and number of aromatic substituents crucial for activity, the requirement for symmetrical systems (1 vs 2), and the importance of the length of the central alkane spacer arm between the two amide moieties present in bis-tyrphostin. These libraries were termed library 1 (dimeric compounds) and library 2 (asymmetric, monomeric compounds).

## 2.2 Synthesis of analogue libraries

Simple application of Knoevenagel chemistry and a series of appropriate  $\alpha$ ,  $\omega$ -bisamines rapidly afforded the desired libraries (Scheme 1) in good to excellent yields.



5 **Scheme 1.** Synthesis of library 1. The  $\text{R}^1$ - $\text{R}^5$  substituents and the alkane spacer  $n$  are defined in Table 2 below.

Utilisation of this approach allowed the rapid generation of five discreet sub-libraries within library 1, based upon the length of the alkane spacer arm with  $n = 1-5$ . Initial biological screens for dynamin I GTPase activity were conducted at  $100 \mu\text{M}$ . More promising analogues were then screened across a range of concentrations to determine their  $\text{IC}_{50}$  values (Table 2). Of the 80 analogues synthesized a number of compounds were

10

found to have an  $IC_{50}$  of 100  $\mu$ M or below and exhibited marked inhibition. The  $R_1$  to  $R_5$  substituents are identified in Table 1 below.

### 2.3 *Synthesis of dimeric tyrphostins*

#### 2.3.1 General

5 All starting materials were purchased from Aldrich Chemical Company and Lancaster Synthesis.  $^1H$  and  $^{13}C$  spectra were recorded on a Bruker Advance AMX 300 MHz spectrometer at 300.1315 and 75.4762 MHz respectively. Chemical shifts are relative to TMS as internal standard.

#### 2.3.2 *Synthetic methods*

##### 10 Compound 5a: *2-Cyano-N-[3-(2-cyanoacetyl-amino)-ethyl]-acetamide*

Ethylenediamine (3a) (1.5 g, 25 mmol) and methylcyanoacetate (5 g, 50 mmol) were stirred at room temperature for 2 hours. The resulting white solid was then mixed with 10 mL ethanol and collected by filtration. Recrystallization from ethanol gave a white solid, 6.3 g (81%). mp 182°C (Lit 183°C)<sup>29</sup>.

15  $^1H$  NMR (DMSO): 8.25 (2H, t,  $J$  = 5.5 Hz), 3.56 (4H, s), 3.13 (4H, br s).

$^{13}C$  NMR (DMSO): 162.31, 115.96, 38.41, 25.25.

##### Compound 5b: *2-Cyano-N-[3-(2-cyanoacetyl-amino)-propyl]-acetamide*

Propanediamine (3b) (2.2 g, 30 mmol) and methylcyanoacetate (6.4 g, 65 mmol) were stirred at room temp for two hours. The resulting white solid was then mixed with 20 mL  
20 of ethanol and collected by filtration. Recrystallization from ethanol gave 4.995 g of white solid (81%). mp 146°C (Lit 148 °C)<sup>29</sup>

$^1\text{H}$  NMR (DMSO): 8.21 (2H, t,  $J=5.5$  Hz), 3.59 (4H, s), 3.07 (4H, q,  $J=6.7$  Hz), 1.53 (2H, quin,  $J=6.7$  Hz).

$^{13}\text{C}$  NMR (DMSO): 162.45, 116.64, 39.28, 28.90, 25.67.

Compound 5c: *2-Cyano-N-[3-(2-cyanoacetylamino)-butyl]-acetamide*

- 5 1,4-diaminobutane (3c) (3 g, 34 mmol) and methylcyanoacetate (7 g, 70 mmol) were stirred at room temp for two hours after which time a white solid was formed. The solid was then mixed with ethanol (10 mL) and collected by filtration. Recrystallization from ethanol gave a white solid, 5.995 g (78%). mp 145°C (Lit 145°C)

$^1\text{H}$  NMR (DMSO): 8.15 (2H, t,  $J=5.5$  Hz), 3.56 (4H, s), 3.05 (4H, br s), 1.38 (4H, br s)

- 10  $^{13}\text{C}$  NMR (DMSO): 161.84, 116.09, 38.63, 26.07, 25.17.

Compound 5d: *2-Cyano-N-[3-(2-cyanoacetylamino)-pentyl]-acetamide*

- 15 1,5-diaminopentane (3d) (2 g, 20 mmol) and methylcyanoacetate (3.9 g, 40 mmol) were stirred at room temp for two hours after which time a white solid was formed. The solid was then mixed with ethanol (10 mL) and collected by filtration. Recrystallization from ethanol gave a white solid, 4.62 g (98%). mp 125°C (Lit 125°C)

$^1\text{H}$  NMR (DMSO): 8.14 (2H, t,  $J=5.4$  Hz), 3.55 (s, 4H), 3.03 (4H, q,  $J=6.4$  Hz), 1.39 (4H, quin,  $J=7$  Hz), 1.23 (2H, quin,  $J=7$  Hz).

$^{13}\text{C}$  NMR (DMSO): 161.79, 116.11, 38.84, 28.26, 25.17, 23.43.

Compound 5e: *2-Cyano-N-[3-(2-cyanoacetylamino)-hexyl]-acetamide*

- 20 1,6 diaminohehexane (3e) (3 g, 26 mmol) and methylcyanoacetate (6 g, 60 mmol) were stirred at room temp for 2 hours after which time a white solid was formed. The solid was then mixed with ethanol (10 mL) and collected by filtration. Recrystallization from ethanol gave a white solid, 6.2 g (95%). mp 141°C (Lit 140 °C)

<sup>1</sup>H NMR (DMSO): 8.15 (2H, t, *J* = 5.5 Hz), 3.56 (4H, s), 3.04 (4H, q, *J* = 6.1 Hz), 1.37 (4H, quin, *J* = 5.9 Hz), 1.24 (4H, br s).

<sup>13</sup>C NMR (DMSO): 161.76, 116.12, 38.85, 28.58, 25.82, 25.16.

Compound 9: 2-Cyano-*N*-{3-[2-cyano-3-(3,4-dihydroxyphenyl)-acryloylamino]-ethyl}-3-(3,4-dihydroxyphenyl)-acrylamide

2-Cyano-*N*-[3-(2-cyano-acetylamino)-ethyl]-acetamide (5a) (0.3 g, 1.5 mmol), 3,4-dihydroxybenzaldehyde (0.42 g, 3 mmol), 3 drops of piperidine and ethanol (10 mL) were refluxed for 2 hours. Cooling, filtering and washing with cold ether (10 mL) gave a yellow-green solid, 0.54 g (81%). mp 290°C (Lit 295°C)

<sup>1</sup>H NMR (DMSO): 8.32 (2H, t, *J* = 5.5 Hz), 7.92 (2H, s), 7.53 (2H, d, *J* = 2.1 Hz), 7.25 (2H, dd, *J* = 8.2, 2.1 Hz), 6.85 (2H, d, *J* = 2.1 Hz), 3.45 (4H, br s).

<sup>13</sup>C NMR (DMSO): 162.50, 151.63, 161.61, 146.22, 125.76, 123.45, 117.65, 116.53, 116.31, 100.85, 39.60.

Compound 10: 2-Cyano-*N*-{3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-ethyl}-3-(3,4,5-trihydroxyphenyl)-acrylamide

2-Cyano-*N*-[3-(2-cyano-acetylamino)-ethyl]-acetamide (5a) (0.056 g, 0.3 mmol), 3,4,5-trihydroxybenzaldehyde (0.1 g, 0.65 mmol) and 1 drop piperidine and ethanol (2 mL) were refluxed for 1 hour. Cooling, filtering and washing with cold ethanol (10 mL) gave an orange solid, 0.11 g (82%). mp >300°C

<sup>1</sup>H NMR (DMSO): 8.29 (2H, t, *J* = 5.5 Hz), 7.79 (2H, s), 6.99 (4H, s), 3.32 (4H, br s).

<sup>13</sup>C NMR (DMSO): 162.15, 150.7, 145.96, 140.24, 121.26, 117.30, 109.97, 99.76, 39.40.

Compound 11: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxy-4-methoxyphenyl)-acryloylamino]-ethyl]-3-(3,4-dihydroxy-5-methoxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyano-acetylamino)-ethyl]-acetamide (5a) (0.06 g, 3 mmol), 3,4-dihydroxy-5-methoxybenzaldehyde (0.1 g, 0.6 mmol), 1 drop of piperidine and 2 mL of ethanol were  
5 refluxed for 2 hours. Cooling, filtering and washing with cold ethanol (5 mL) gave an orange solid, 0.101 g (66%). mp 274°C

<sup>1</sup>H NMR (DMSO): 8.34 (2H, t, *J* = 5.5 Hz), 7.93 (1H, s), 7.20 (2H, d, *J* = 1.92 Hz), 7.13 (2H, d, *J* = 1.92 Hz), 3.77 (6H, s), 3.35 (4H, br s).

<sup>13</sup>C NMR (DMSO): 161.90, 150.85, 148.03, 145.83, 139.90, 121.76, 117.20, 111.09, 107.20,  
10 100.83.

Compound 22: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxyphenyl)-acryloylamino]-propyl]-3-(3,4-dihydroxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetylamino)-propyl]-acetamide (5b) (0.3 g 1.4 mmol), (0.4 g, 2.8 mmol) 3,4-dihydroxybenzaldehyde, 3 drops of piperidine and 10 mL of ethanol were  
15 refluxed for 2 hours. Cooling, filtering and washing with cold ether (10 mL) gave a yellow green solid, 0.55 g (85%). mp 274°C (Lit 277°C)

<sup>1</sup>H NMR (DMSO): 8.24 (2H, t, *J* = 5.5 Hz), 7.92 (s, 2H), 7.52 (2H, d, *J* = 2.1 Hz), 7.26 (2H, dd, *J* = 8.2, 2.1 Hz), 6.85 (2H, d, *J* = 8.2 Hz), 3.23 (4H, q, *J* = 6 Hz), 1.70 (2H, quin, *J* = 6.7 Hz).

<sup>13</sup>C NMR (DMSO): 161.50, 150.60, 150.50, 125.10, 123.21, 117.10, 116.00, 115.80, 100.50, 37.27,  
20 28.82.

Compound 23: *2-Cyano-N-[3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-propyl]-3-(3,4,5-trihydroxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-propyl]-acetamide (5b) (0.06 g 0.29 mmol), 3,4,5-trihydroxybenzaldehyde (0.1 g, 0.58 mmol), 1 drop of piperidine and ethanol (10 mL) were refluxed for 2 hours. Cooling, filtering and washing with cold ethanol (10 mL) gave an orange solid, 0.097 g (70%). Mp >300°C (Lit >300°C).

- 5 <sup>1</sup>H NMR (DMSO): 8.18 (2H, t, *J* = 5.5 Hz), 7.78 (2H, s), 6.99 (4H, s), 3.21 (4H, q, *J* = 6.8 Hz), 1.68 (2H, quin, *J* = 6.8 Hz).

<sup>13</sup>C NMR (DMSO): 161.80, 150.70, 145.95, 140.30, 121.22, 117.30, 109.90, 99.50, 38.20, 28.90.

Compound 24: 2-Cyano-*N*-[3-[2-cyano-3-(3,4-dihydroxy-5-methoxyphenyl)-acryloylamino]-propyl]-3-(3,4-dihydroxy-5-methoxyphenyl)-acrylamide

- 10 2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-propyl]-acetamide (5b) (0.3 g 1.4 mmol), 0.44 g 3,4-dihydroxy-4-methoxybenzaldehyde, 3 drops of piperidine and ethanol (10 mL) were refluxed for 2 hours. Cooling, filtering and washing with cold ethanol (5 mL) gave an orange solid, 0.31 g (42%). mp >300°C

- 15 <sup>1</sup>H NMR (DMSO): 8.35 (2H, t, *J* = 5.4 Hz), 7.95 (2H, s), 7.21 (2H, d, *J* = 1.9 Hz), 7.12 (2H, d, *J* = 1.9 Hz), 3.21 (4H, q, *J* = 6.8 Hz), 1.71 (2H, quin, *J* = 6.8 Hz).

<sup>13</sup>C NMR (DMSO): 161.30, 150.61, 147.20, 145.30, 121.04, 117.60, 110.60, 107.65, 98.71, 38.35, 28.88.

Compound 35: 2-Cyano-*N*-[3-[2-cyano-3-(3,4-dihydroxyphenyl)-acryloylamino]-butyl]-3-(3,4-dihydroxyphenyl)-acrylamide

- 20 2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-butyl]-acetamide (5c) (0.3 g 1.35 mmol), 3,4-dihydroxybenzaldehyde (0.37 g, 2.7 mmol), 3 drops of piperidine and ethanol (10 mL) were

refluxed for 2 hours. Cooling, filtering and washing with cold ether (10 mL) gave a yellow solid, 0.61 g (97%). mp 281°C (Lit 283 °C)

<sup>1</sup>H NMR (DMSO): 8.25 (2H, t, *J*=5.5 Hz), 7.91 (2H, s), 7.53 (2H, d, *J*=1.9 Hz), 7.26 (2H, dd, *J*=8.3, 1.9 Hz), 6.85 (2H, d, *J*=8.3 Hz), 3.20 (4H, br s), 1.49 (4H, br s).

5 <sup>13</sup>C NMR (DMSO): 161.52, 150.86, 150.42, 145.65, 125.23, 123.09, 117.20, 115.81, 100.51, 39.31.

Compound 36: 2-Cyano-*N*-{3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-butyl}-3-(3,4,5-trihydroxyphenyl)-acrylamide

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-butyl]-acetamide (5c) (0.065 g, 0.3 mmol), 3,4,5-trihydroxybenzaldehyde (0.1 g, 0.6 mmol), 1 drop of piperidine and ethanol (2 mL) were  
10 refluxed for 1 hour. Cooling, filtering and washing w 2-Cyano-*N*-[3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-butyl]-3-(3,4,5-trihydroxyphenyl)-acrylamide ith cold ether (5 mL) gave a yellow solid, 0.121 g (82%). mp >300°C (Lit >310°C)<sup>29</sup>

<sup>1</sup>H NMR (DMSO): 8.16 (2H, t, *J*=5.5 Hz), 7.78 (2H, s), 6.98 (4H, s), 3.19 (4H, br s), 1.48 (4H, br s).

15 <sup>13</sup>C NMR (DMSO): 161.70, 150.56, 145.90, 140.20, 121.30, 117.30, 109.90, 99.80, 39.26, 26.37.

Compound 37: 2-Cyano-*N*-{3-[2-cyano-3-(3,4-dihydroxy-5-methoxyphenyl)-acryloylamino]-butyl}-3-(3,4-dihydroxy-5-methoxyphenyl)-acrylamide

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-butyl]-acetamide (5c) (0.065 g, 0.3 mmol), 3,4-dihydroxy-5-methoxybenzaldehyde (0.1 g, 0.6 mmol), 1 drop of piperidine and ethanol (2  
20 mL) were refluxed for 1 hour. Cooling, filtering and washing with cold ether (5 mL) gave a yellow solid, 0.110 g (70%). mp >300°C



$^1\text{H}$  NMR (DMSO): 8.09 (2H, t,  $J=5.5$  Hz), 7.86 (2H, s), 7.18 (2H, d,  $J=1.9$  Hz), 7.10 (2H, d,  $J=1.9$  Hz), 3.75 (6H, s), 3.19 (4H, br s), 1.48 (4H, br s).

$^{13}\text{C}$  NMR (DMSO): 161.71, 150.23, 148.70, 146.24, 120.25, 117.51, 109.50, 106.80, 98.81, 55.76, 39.31, 26.64.

5 Compound 48: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxyphenyl)-acryloylamino]-pentyl]-3-(3,4-dihydroxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-pentyl]-acetamide (5d) (0.2 g, 0.85 mmol), 3,4-dihydroxybenzaldehyde (0.23g, 1.7 mmol), 3 drops of piperidine and 7 mL ethanol were refluxed for 2 hours. Cooling, filtering and washing with cold ether (10 mL) gave a yellow  
10 solid, 0.36 g (90%). mp 252°C (Lit 248°C)<sup>29</sup>

$^1\text{H}$  NMR (DMSO): 8.15 (2H, t,  $J=5.5$  Hz), 7.85 (2H, s), 7.50 (2H, d,  $J=2.1$  Hz), 7.20 (2H, dd,  $J=8.5\text{Hz}, 2\text{ Hz}$ ), 6.75 (2H, d,  $J=8.5\text{Hz}$ ), 3.16 (4H, q,  $J=6.2$  Hz), 1.50 (4H, quin,  $J=7.1$  Hz), 1.28 (2H, quin,  $J=6.9$  Hz).

$^{13}\text{C}$  NMR (DMSO): 161.85, 153.88, 150.34, 146.28, 126.16, 121.47, 117.70, 115.71, 114.65, 98.40,  
15 39.46, 28.63, 23.73.

Compound 49: *2-Cyano-N-[3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-pentyl]-3-(3,4,5-trihydroxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-pentyl]-acetamide (5d) (0.068 g, 0.29 mmol), (0.1 g, 0.58 mmol) 3,4,5-trihydroxybenzaldehyde (0.1g, 0.58 mmol), 1 drop of piperidine and ethanol (2  
20 mL) were refluxed for 1 hour. Cooling, filtering and washing with cold ether (5 mL) gave a yellow solid, 0.123 g (83%). mp >300°C<sup>32</sup>

<sup>1</sup>H NMR (DMSO): 8.12 (2H, t, *J* = 5.5 Hz), 7.76 (2H, s), 6.98 (4H, s), 3.16 (4H, br s), 1.50 (4H, quin, *J* = 6.8 Hz), 1.28 (2H, quin, *J* = 6.7 Hz).

<sup>13</sup>C NMR (DMSO): 161.80, 150.49, 146.11, 146.01, 141.25, 120.69, 117.53, 109.90, 99.12, 39.47, 28.62, 22.30.

5    Compound 50: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxy-5-methoxyphenyl)-acryloylamino]-pentyl]-3-(3,4-dihydroxy-5-methoxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-pentyl]-acetamide (5d) (0.069 g, 0.29 mmol) 3,4-dihydroxy-5-methoxybenzaldehyde (0.1g, 0.58 mmol), 1 drop of piperidine and ethanol (2 mL) were refluxed for 1 hour. Cooling, filtering and washing with cold ether (5 mL) gave a  
10    yellow solid, 0.126 g (81%). mp 256°C

<sup>1</sup>H NMR (DMSO): 8.09 (2H, t, *J* = 5.5 Hz), 7.86 (2H, s), 7.18 (2H, d, *J* = 2 Hz), 7.10 (2H, d, *J* = 2 Hz) 3.75 (6H, s), 3.17 (4H, br s), 1.50 (4H, quin, *J* = 6.8 Hz), 1.28 (4H, quin, *J* = 6.9 Hz).

<sup>13</sup>C NMR (DMSO): 161.81, 150.50, 148.00, 146.20, 120.05, 117.81, 110.50, 107.80, 98.71, 55.71, 39.41, 28.64, 22.47.

15    Compound 61: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxyphenyl)-acryloylamino]-hexyl]-3-(3,4-dihydroxyphenyl)-acrylamide*

2-Cyano-*N*-[3-(2-cyanoacetyl-amino)-hexyl]-acetamide (5e) (0.3 g, 1.2 mmol), 3,4-dihydroxybenzaldehyde (0.33 g, 2.4 mmol), 3 drops of piperidine and 10 mL ethanol were refluxed for 2 hours. Cooling, filtering and washing with cold ether (10 mL) gave a yellow  
20    solid, 0.52 g (89%). mp 263°C (Lit 260 °C)

<sup>1</sup>H NMR (DMSO): 8.18 (2H, t, *J* = 5.5 Hz), 7.89 (2H, s), 7.51 (2H, d, *J* = 2 Hz), 7.24 (2H, dd, *J* = 2 Hz, 8.3 Hz), 6.83 (2H, d, *J* = 8.3 Hz), 3.17 (4H, q, *J* = 6.1 Hz), 1.47 (4H, quin, *J* = 6.1 Hz), 1.28 (4H, br s).

<sup>13</sup>C NMR (DMSO): 161.52, 151.18, 150.30, 145.72, 125.22, 122.91, 117.24, 115.80, 115.72,  
5 100.38, 39.48, 28.78, 25.98.

Compound 62: *2-Cyano-N-[3-[2-cyano-3-(3,4,5-trihydroxyphenyl)-acryloylamino]-hexyl]-3-(3,4,5-trihydroxyphenyl)-acrylamide*

2-Cyano-*N*[3-(2-cyanoacetyl-amino)-hexyl]-acetamide (5e) (0.073 g, 0.29 mmol), 3,4,5-trihydroxybenzaldehyde (0.1 g, 0.58 mmol), 1 drop of piperidine and ethanol (2 mL) were  
10 refluxed for 1 hour. Cooling, filtering and washing with cold ether (5 mL) gave a yellow solid, 0.1 g (67%). mp >300°C

<sup>1</sup>H NMR (DMSO): 8.11 (2H, t, *J* = 5.5 Hz), 7.76 (2H, s), 6.98 (4H, s), 3.16 (4H, br s), 1.47 (4H, quin, *J* = 6.1 Hz), 1.28 (4H, br s).

<sup>13</sup>C NMR (DMSO): 161.80, 150.46, 145.99, 141.19, 120.71, 117.53, 109.89, 99.15, 39.68, 28.84,  
15 26.01.

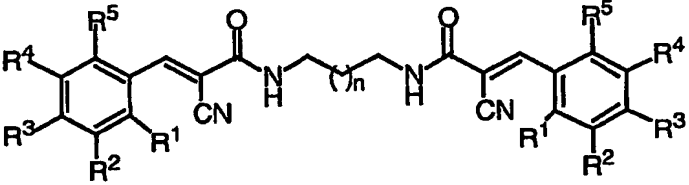
Compound 63: *2-Cyano-N-[3-[2-cyano-3-(3,4-dihydroxy-5-methoxyphenyl)-acryloylamino]-hexyl]-3-(3,4-dihydroxy-5-methoxyphenyl)-acrylamide*

2-Cyano-*N*[3-(2-cyanoacetyl-amino)-hexyl]-acetamide (5e) (0.069 g, 0.28 mmol), 3,4-dihydroxy-5-methoxybenzaldehyde (0.1 g, 0.56 mmol), 1 drop of piperidine and ethanol  
20 (2 mL) were refluxed for 1 hour. Cooling, filtering and washing with cold ether (5 mL) gave a yellow solid, 0.132 g (86%). mp 243°C

$^1\text{H}$  NMR (DMSO): 8.17 (2H, t,  $J = 5.5$  Hz), 7.89 (2H, s), 7.19 (2H, d,  $J = 1.6$  Hz), 7.13 (2H, d,  $J = 1.6$  Hz), 3.77 (6H, s), 3.17 (4H, br s), 1.48 (4H, quin,  $J = 6.1$  Hz), 1.29 (4H, br s).

$^{13}\text{C}$  NMR (DMSO): 161.80, 150.49, 146.11, 146.01, 141.25, 120.69, 117.53, 109.90, 99.12, 39.47, 28.62, 22.30.

5 TABLE 1: Effect of library 1 (dimeric) compounds on dynamin I GTPase activity.



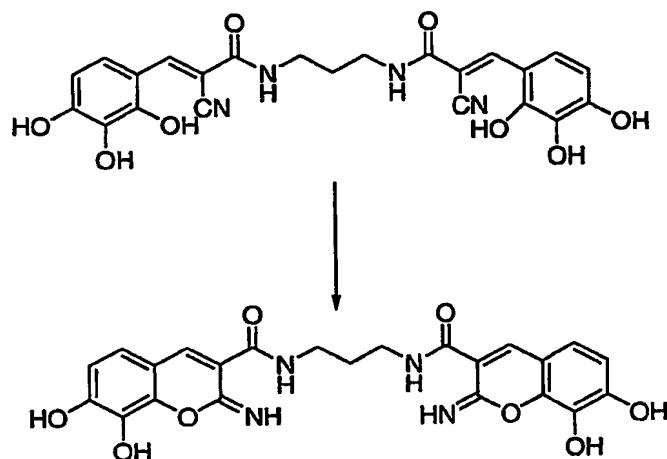
Compound	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	n	IC <sub>50</sub> (μM) <sup>a</sup>
9	H	H	OH	OH	H	1	5.1 ± 0.6
10	H	OH	OH	OH	H	1	1.7 ± 0.2
11	H	OMe	OH	OH	H	1	9 ± 3
22	H	H	OH	OH		2	1.7 ± 0.5
23	H	OH	OH	OH		2	1.7 ± 0.2
24	H	OMe	OH	OH		2	5 ± 1
35	H	H	OH	OH		3	3.2 ± 1
36	H	OH	OH	OH		3	2.1 ± 0.2
37	H	OMe	OH	OH		3	8 ± 0.15
48	H	H	OH	OH		4	5 ± 1.4
49	H	OH	OH	OH		4	1.7 ± 0.4
50	H	OMe	OH	OH		4	8 ± 0.15
61	H	H	OH	OH		5	26 ± 15
62	H	OH	OH	OH		5	6 ± 2
63	H	OMe	OH	OH		5	80 ± 4

Mono-substituted aromatic compounds containing no substitutions, or single substituent such as a single -OH (eg, R<sub>1</sub> or R<sub>2</sub> is OH), -Cl (R<sub>2</sub> or R<sub>4</sub> is Cl), -OMe (R<sub>2</sub> or R<sub>3</sub> is OMe), or -COOH (R<sub>3</sub> is COOH) showed no dynamin inhibition. Introduction of a second

oxygen-bearing substituent had a pronounced effect. The 3,4-di-OH (11,  $IC_{50} = 5.1 \pm 0.6 \mu M$ ) displayed similar potency to compound 1, namely bis-tyrphostin (2,3-di-OH). The 3,4,5-tri-substituted aromatic compound (10) also had equivalent potency to 1. Essentially the same trend was observed for each series of different chain length compounds.

- 5 Alkane spacer chain elongation had little effect on potency until  $n > 3$ . For example, chain extended analogues of 9 ( $n = 0$ ), i.e. 22 ( $n = 1$ ), 35 ( $n = 2$ ), 48 ( $n = 3$ ), and 61 ( $n = 4$ ) displayed  $IC_{50}$  values of  $5.1 \pm 0.6$ ,  $1.7 \pm 0.2$ ,  $3.2 \pm 1$ ,  $5 \pm 1.4$  and  $26 \pm 1.5 \mu M$ , respectively. Essentially the opposite trend was previously reported for tyrosine kinase inhibition. Whilst examining compounds against EGF receptor tyrosine kinase phosphorylation of a
- 10 poly-GAT substrate, Gazit et al observed that inhibition was independent of chain length (Gazit et al., 1996).

Analogues in which  $R_1$  and the position occupied by the cyanyl group (CN) are cyclised may also be provided. For instance, when  $R_1$  is hydroxy, the hydroxy group can react with cyanyl to form an imminochromene as show in Scheme 2 below.

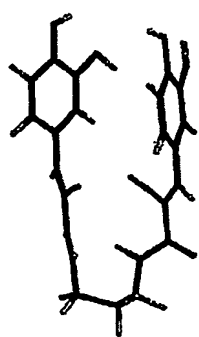
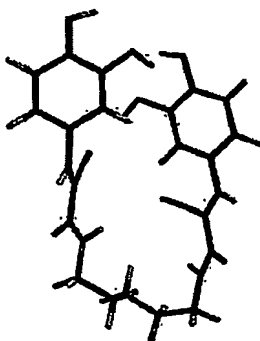
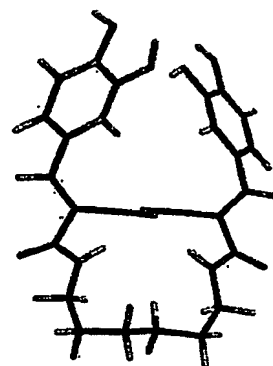
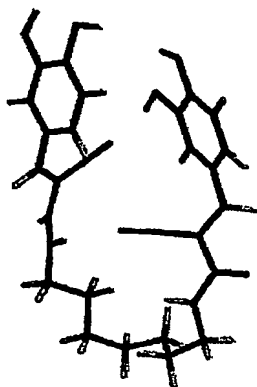
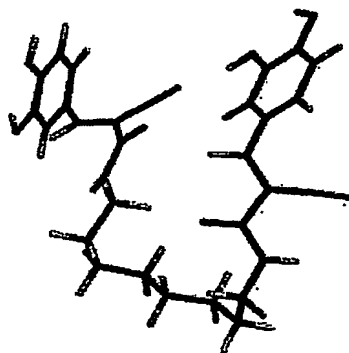
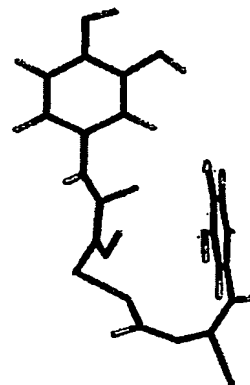


15

**Scheme 2:** Synthesis of an imminochromene analogue of bis-tyrphostin

- To attempt to explain similarities in inhibitory values for the chain-extended analogues of 9, modelling analysis of all 5 alkane spacer analogues was conducted and the
- 20 resulting MacSpartanPro low energy conformer models are shown below. As can be seen, the low energy conformers of all 5 analogues adopt comparable hairpin conformations,

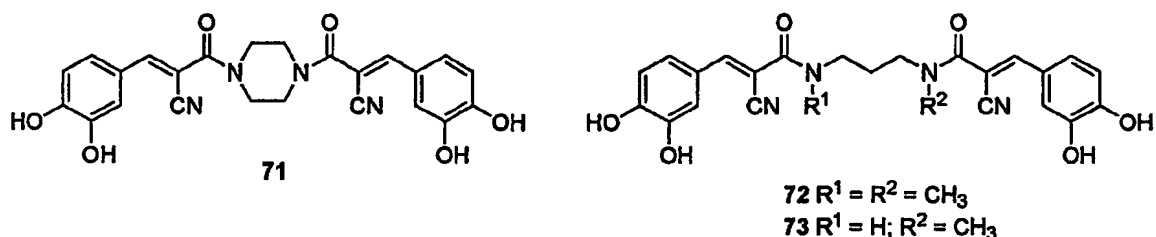
maximizing pi-pi interactions between the terminal phenyl rings. Consequently, increasing the spacer length has limited impact until entropic effects begin to impinge on the relative stability of the hairpin conformation ( $n \geq 5$ ). This contrasts with the effects of dimeric tyrphostins on tyrosine kinase potency which resides in their extended configuration and thus allows them to fit the dimeric intermediate of the EGFR tyrosine kinases.

(a) 9,  $n=0^a$ (b) 22,  $n=1$ (c) 35,  $n=2$ (d) 48,  $n=3$ (e) 61,  $n=1$ 

(f) 121, oxidised S-S form of 2

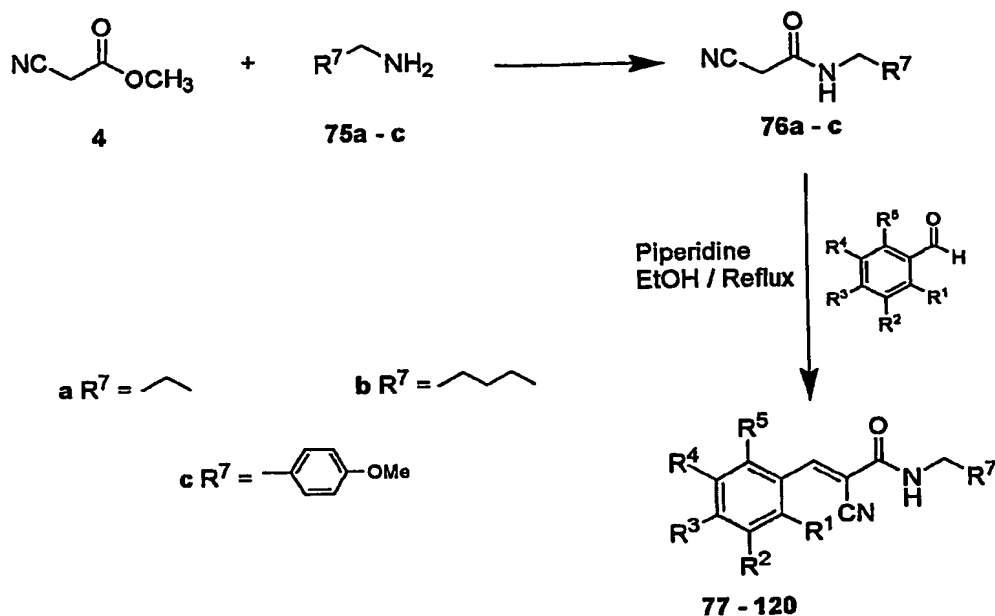
<sup>a</sup>In Table 2, Bis-tyrphostin (1) is also identified as compound 9 which was synthesised according to scheme 1. Thus 1 and 9 are identical.

To explore the potential H-bonding effects associated with 1, compound 71 was developed. This compound has a relatively inflexible piperazine linker of similar overall size to 1. However, it displayed no dynamin inhibition at  $\leq 100 \mu\text{M}$ . Similarly, no inhibitory effect was observed after N-methylating the alkane spacer of 1 to produce N-methyl analogue 72. These observations suggest that the hairpin conformation of dimeric tryphostins is desirable for inhibitory action supporting the modelling observations (hairpin conformation rather than extended chain), and that the amide substituents also play an important role in binding to dynamin.



Structures of compounds 71-73

Having successfully developed a number of  $\mu\text{M}$  potent symmetrical analogues based on bis-tryphostin, modifications of one of the aromatic nuclei were investigated to determine its role in inhibiting dynamin. Accordingly, another compound library based on tryphostin A47 (2) was developed as shown in scheme 3, and the analogues ability to inhibit dynamin I GTPase activity was examined.

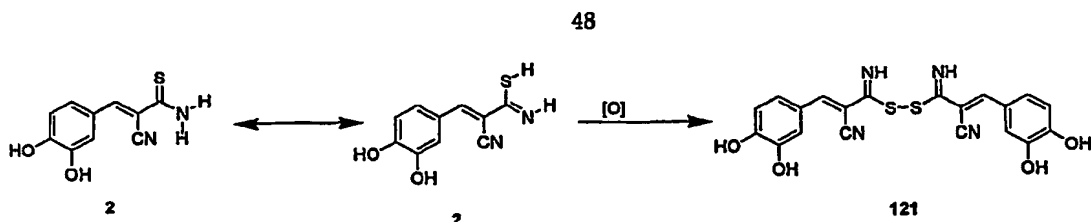


Scheme 3. Synthesis of library 2

Surprisingly, screening of library 2 compounds failed to reveal any with dynamin inhibition  $\leq 100 \mu\text{M}$ . This is more surprising given that the original screening data showed that tyrphostin A47 (2) displayed a dynamin  $\text{IC}_{50} = 70 \mu\text{M}$ .

Closer examination of tyrphostin A47 afforded a potential explanation for the failure to detect inhibitory activity in library 2. That is, the single -S was potentially available for oxidation in solution to the corresponding dimeric structure. Simple tautomerisation followed by oxidation yields the corresponding disulfide species (121) (see Scheme 4). Freshly prepared solutions of 2 showed no inhibitory potency, while stocks kept at room temperature for 24 hrs showed weak potency. The  $\text{IC}_{50}$  of 121 decreased to  $>300 \mu\text{M}$  when the reducing reagent dithiothreitol (2 mM) was included in the dynamin assay medium. Dithiothreitol alone was without effect on dynamin GTPase activity (data not shown). The in situ generation of the dimeric 121 affords a similar low energy conformation with the required key functional groups appropriately disposed to ensure good inhibition of dynamin. A similar sequence of events has been observed for thioindoles which are EGFR tyrosine kinase inhibitors which showed increased activity upon oxidation (Thompson et al., 1993).





Scheme 4

## 2.4 Discussion

The structure-activity relationship of dimeric typhostins against the GTPase enzyme dynamin was evaluated via the synthesis and screening of a library of compounds based upon the lead compounds bis-tyrphostin and typhostin A47. From the results obtained, potent inhibitory activity was found in dimeric tyrphostin compounds containing two aromatic rings with hydroxy groups in the 3,4 positions. Modifications to these compounds can be readily made by altering which functional groups are used to form the linker.

It will be appreciated by persons skilled in the art that numerous variations and/or modifications may be made to the invention as shown in the specific embodiments without departing from the spirit or scope of the invention as broadly described. The preferred embodiments are, therefore, to be considered in all respects illustrative and not restrictive.

Dated this 9<sup>th</sup> day of December 2003

The University of Newcastle Research Associates Limited and  
Children's Medical Research Institute

By their Patent Attorneys  
Blake Dawson Waldron Patent Services

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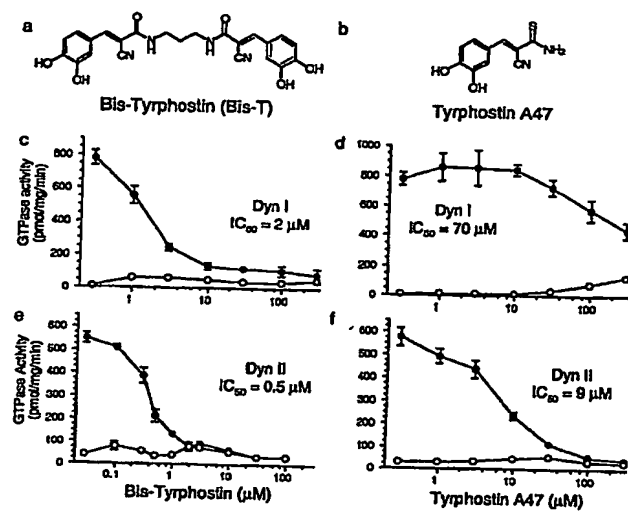
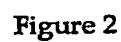


Figure 1



1

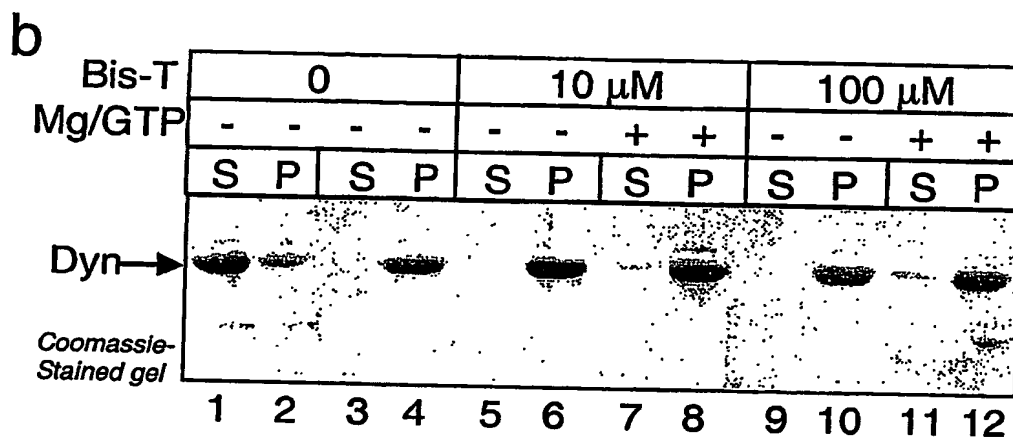
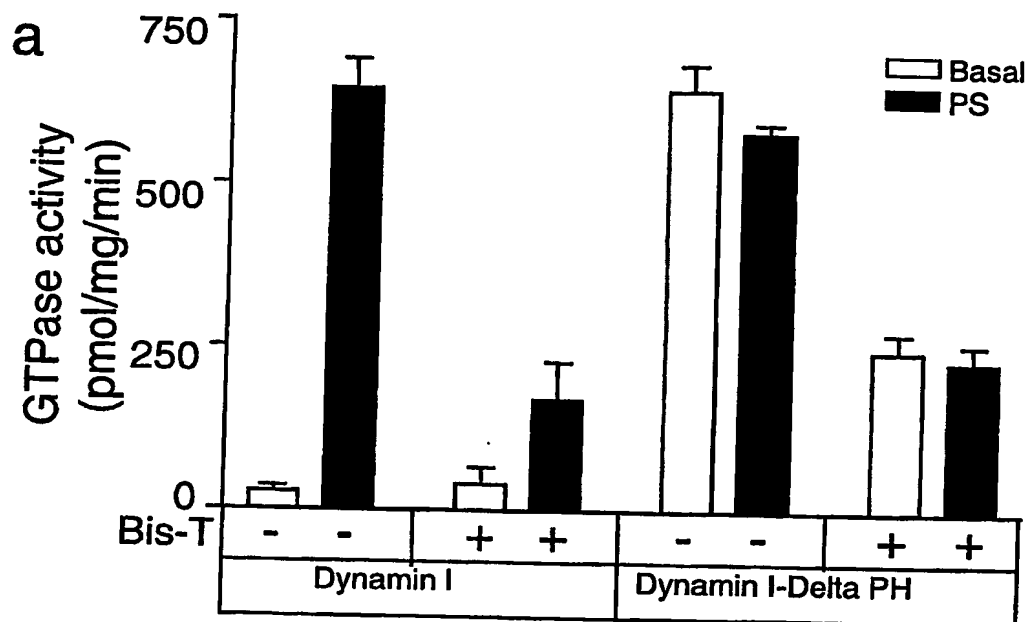


Figure 3



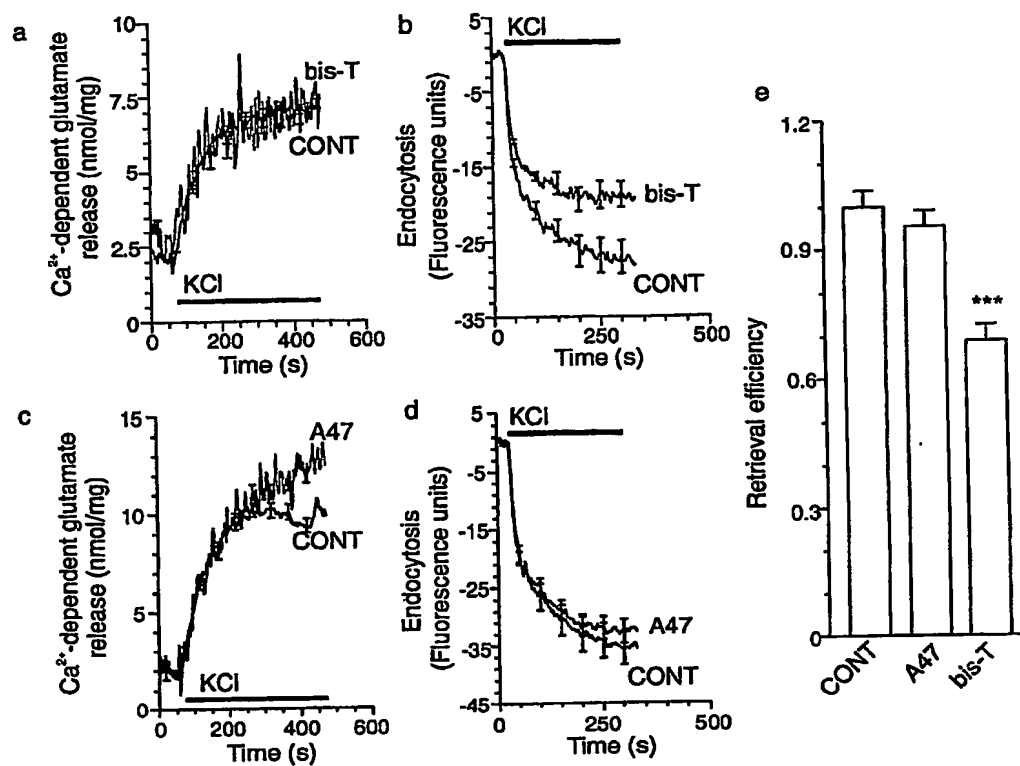


Figure 4

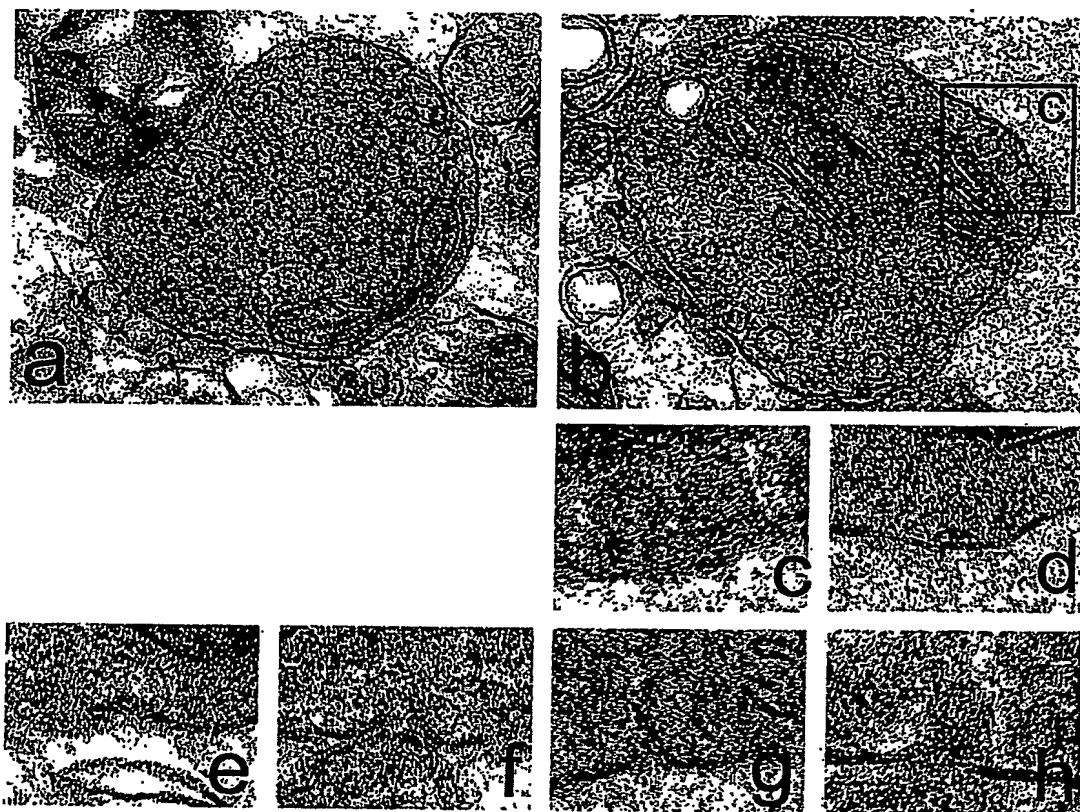


Figure 5

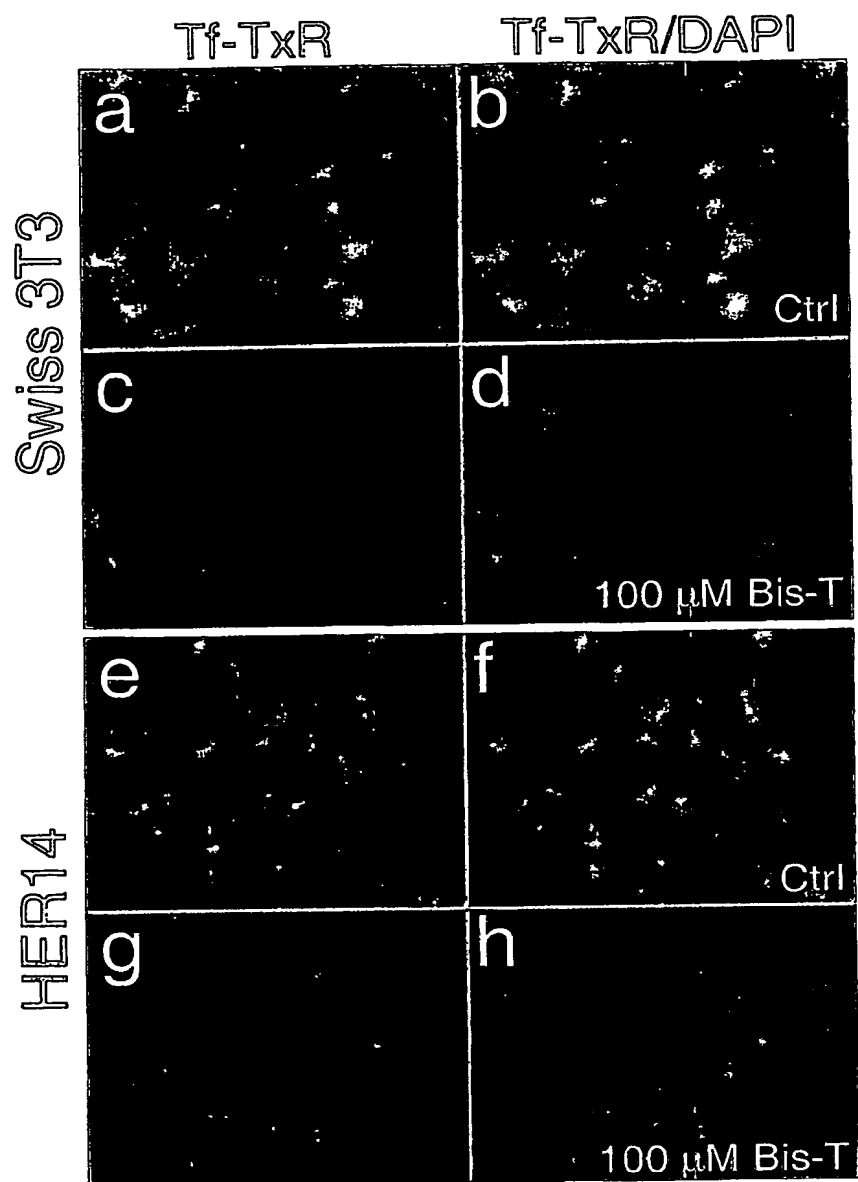


Figure 6

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